

Molecular Nanomagnets

E. Coronado

- Polyoxometalate clusters
- SMMs in different environments



Outline

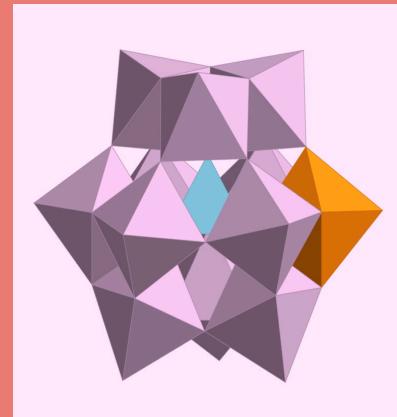
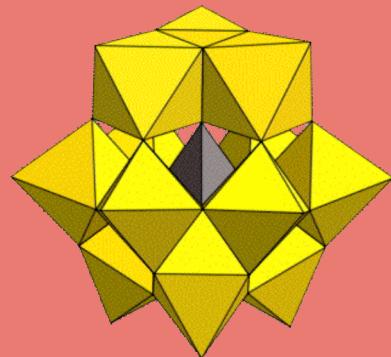
I. Introduction to POMs

II. SMMs and Quantum tunnelling
in lanthanide POMs

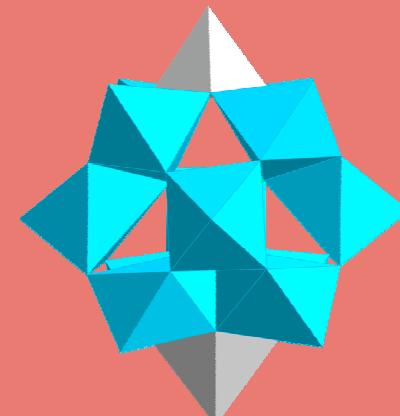
III. SMMs into superconductors

Polyoxometalates : Molecular metal-oxide anions

Keggin



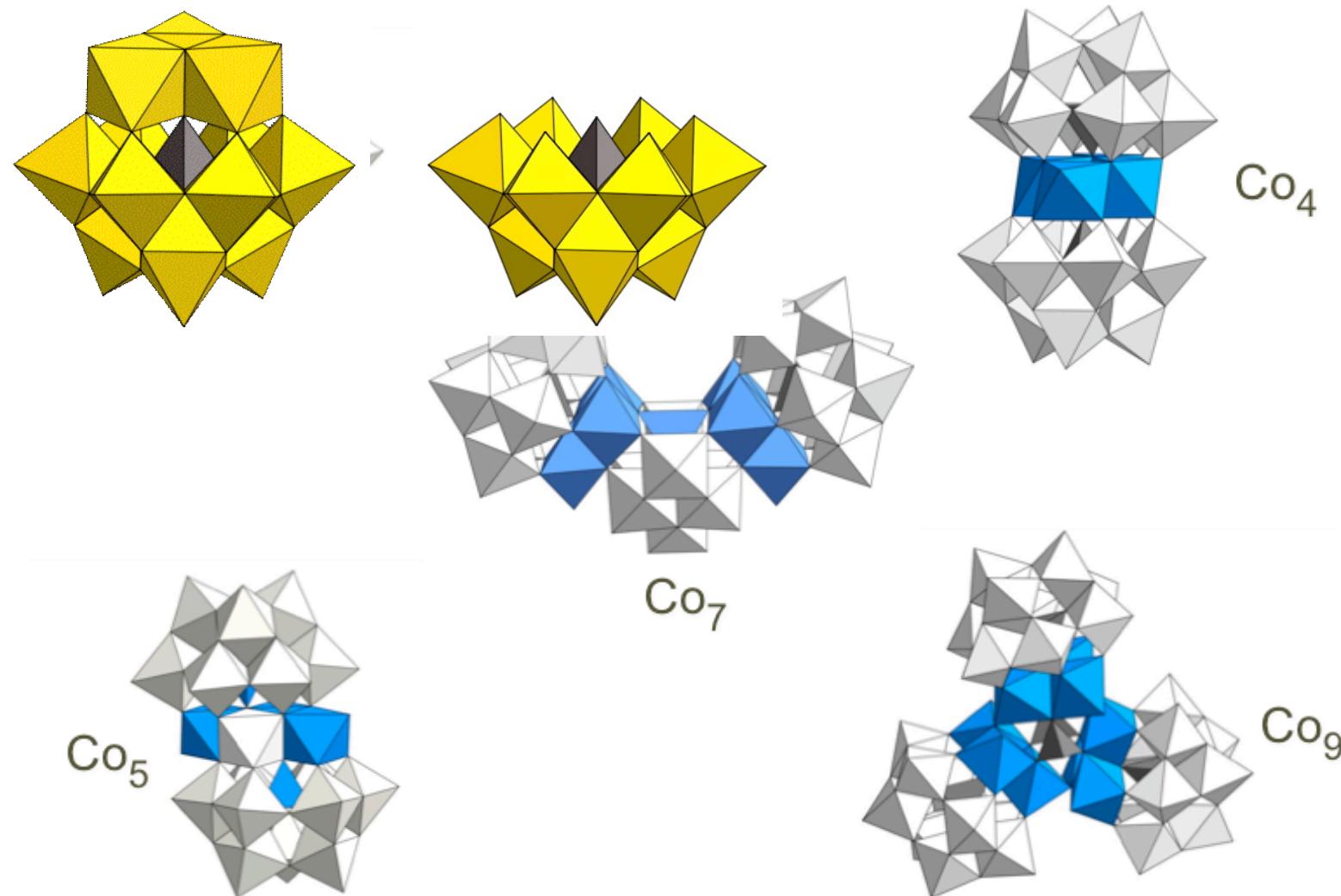
bi-capped
Keggin



Structural/Chemical versatility:
Soluble Metal-oxide cluster anions with defined topologies

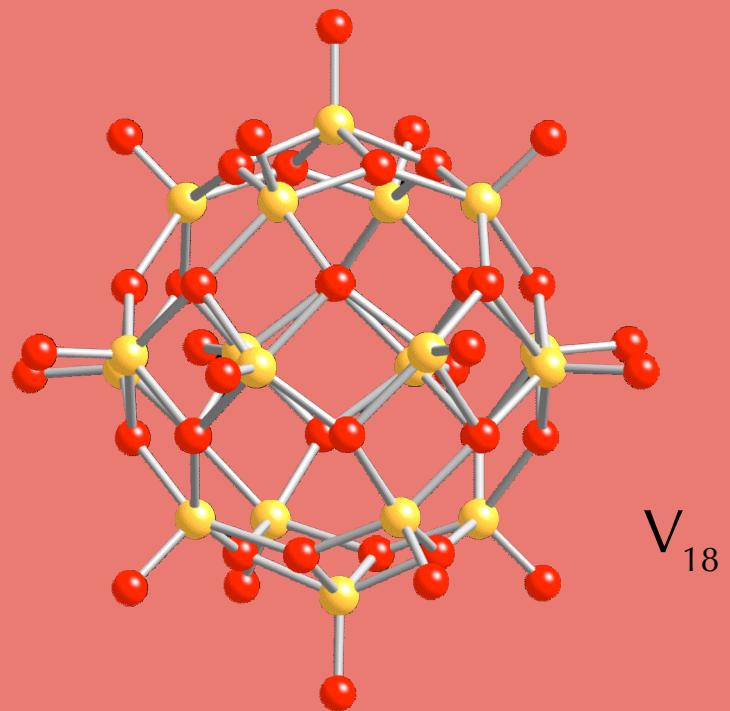
Electronic versatilities:
Magnetic Exchange and/or Electron Transfer
Magnetic anisotropy

Magnetic clusters encapsulated by POMs

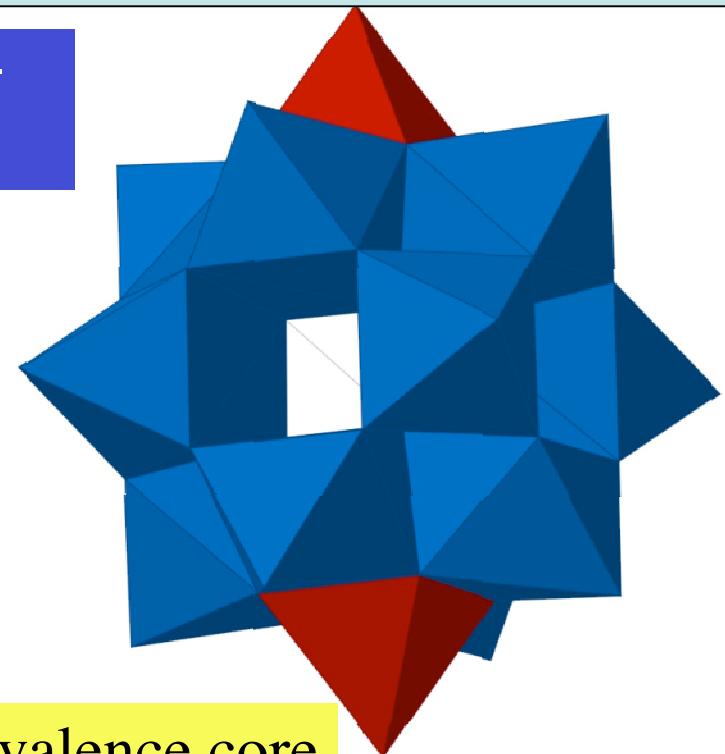
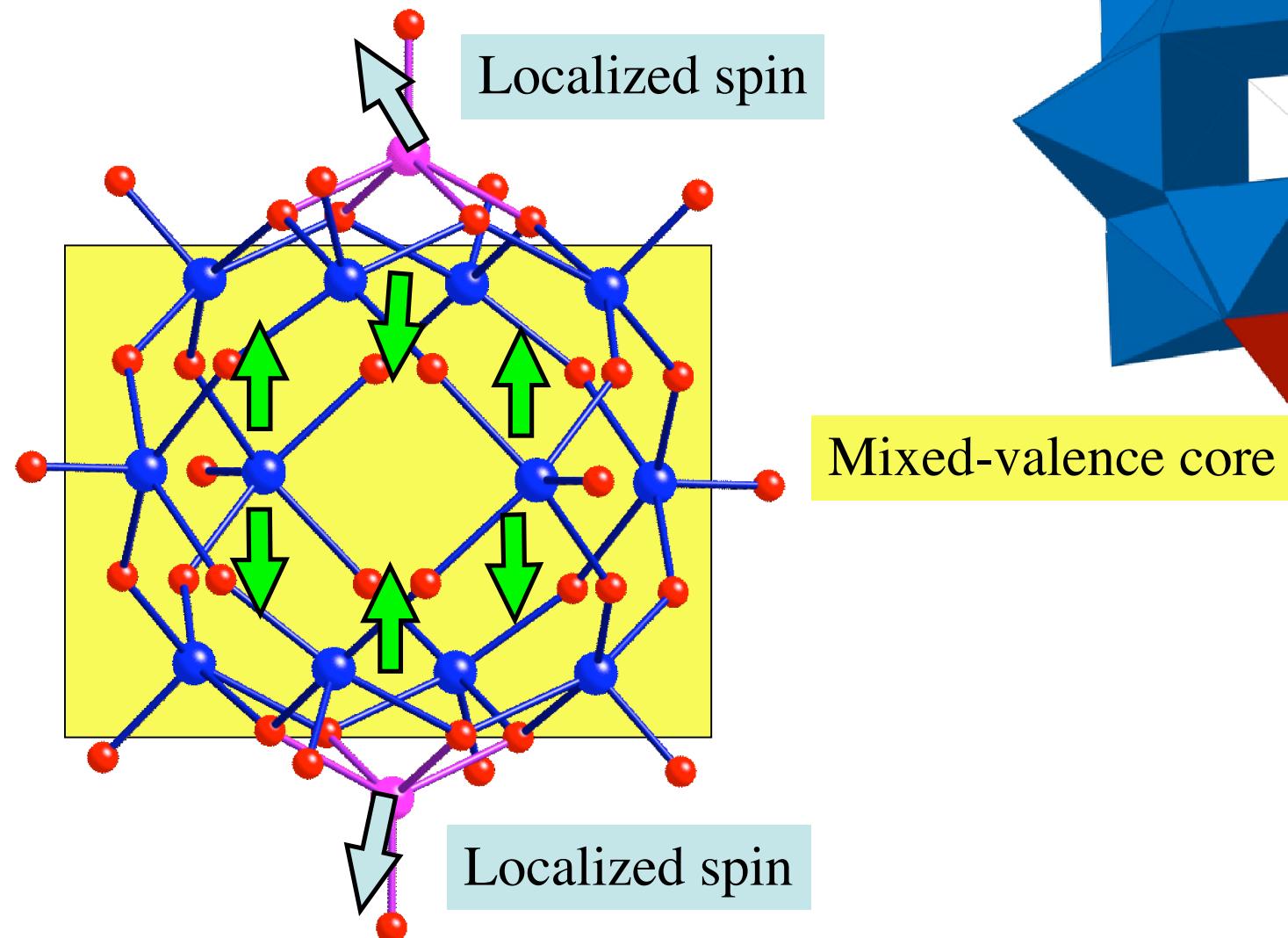


J. M. Clemente-Juan, E. Coronado, *Coord. Chem. Rev.* **1999**, 193-195, 361

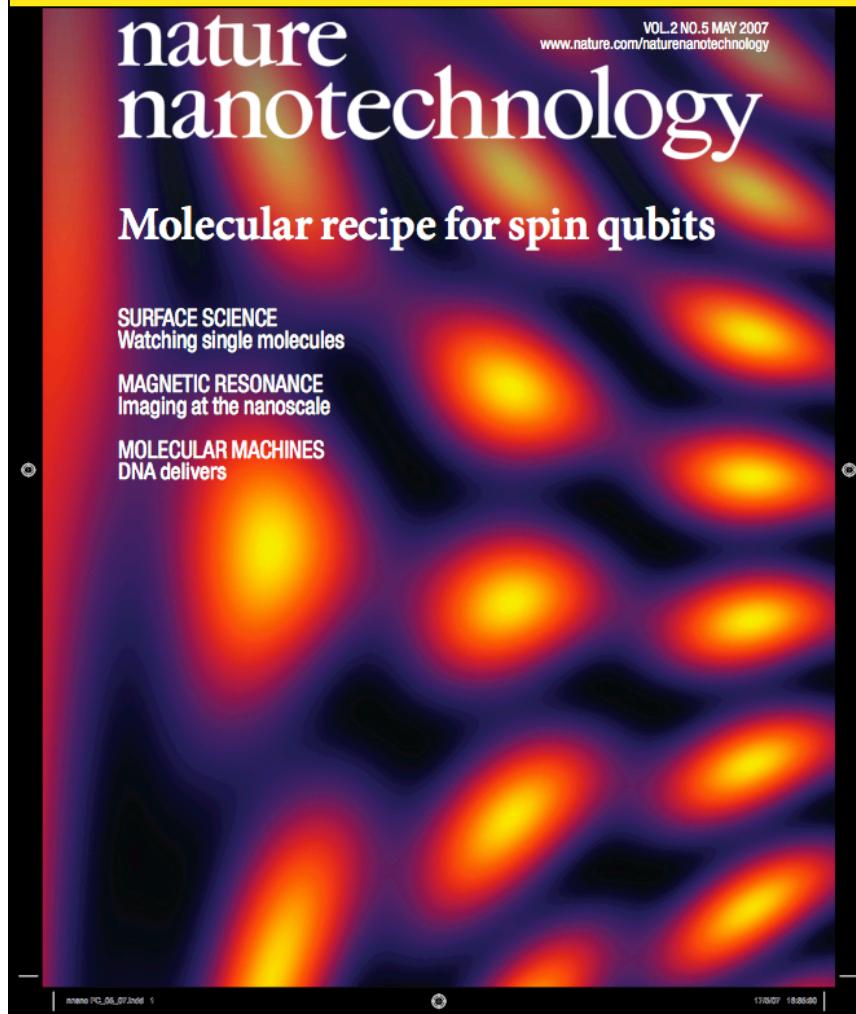
Magnetic clusters of POMs



Magnetic mixed-valence POMs



Applications: Quantum computing

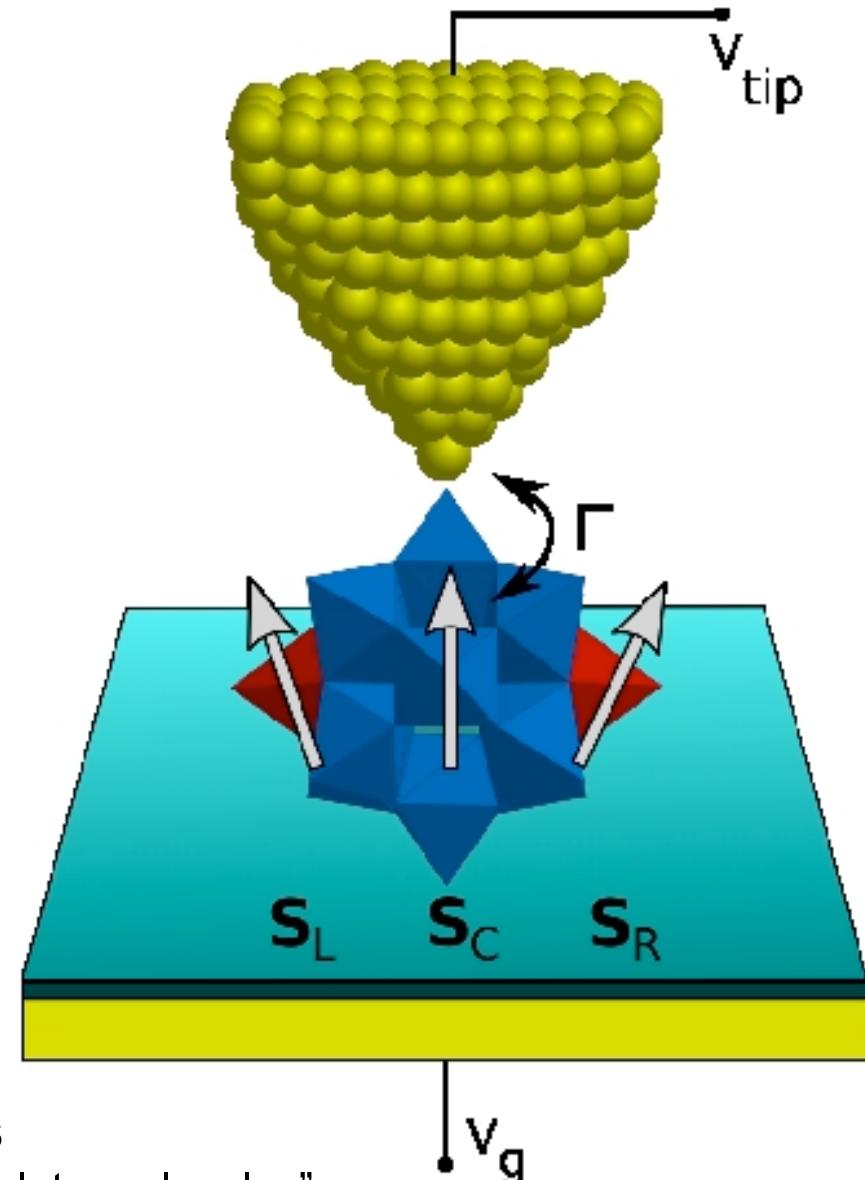


Magnetic Molecular clusters as Qubits

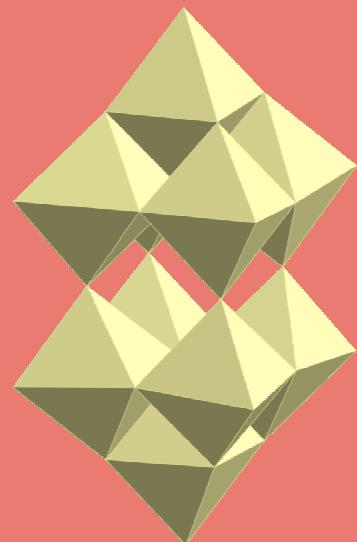
J. Lehmann, A. Gaita-Ariño, E. Coronado, D. Loss

“Spin qubits with electrically gated polyoxometalate molecules”

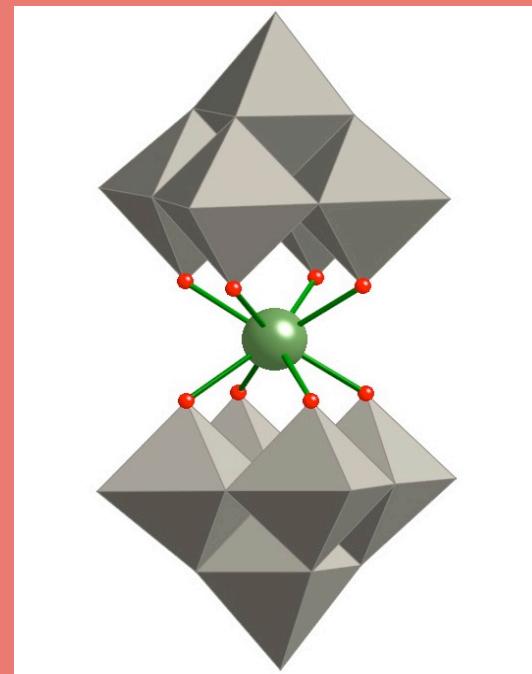
Nature Nanotechnology 2, 312-317 (2007)



Rare-Earths encapsulated by POMs



W_{10}



LnW_{10}

Outline

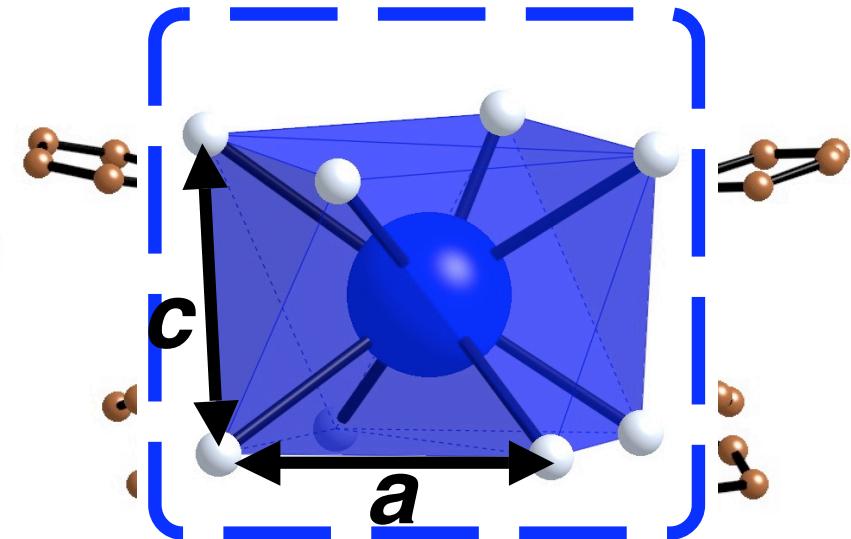
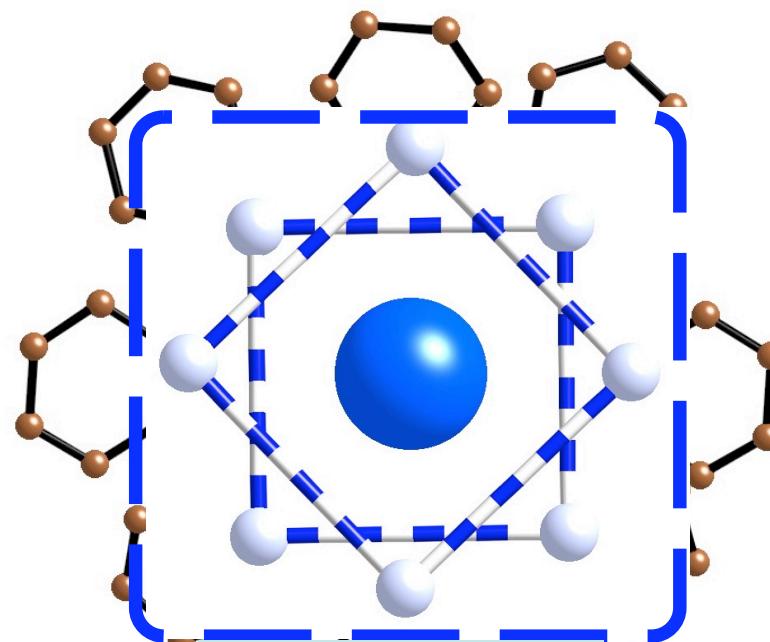
I. Introduction to POMs

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Structural description

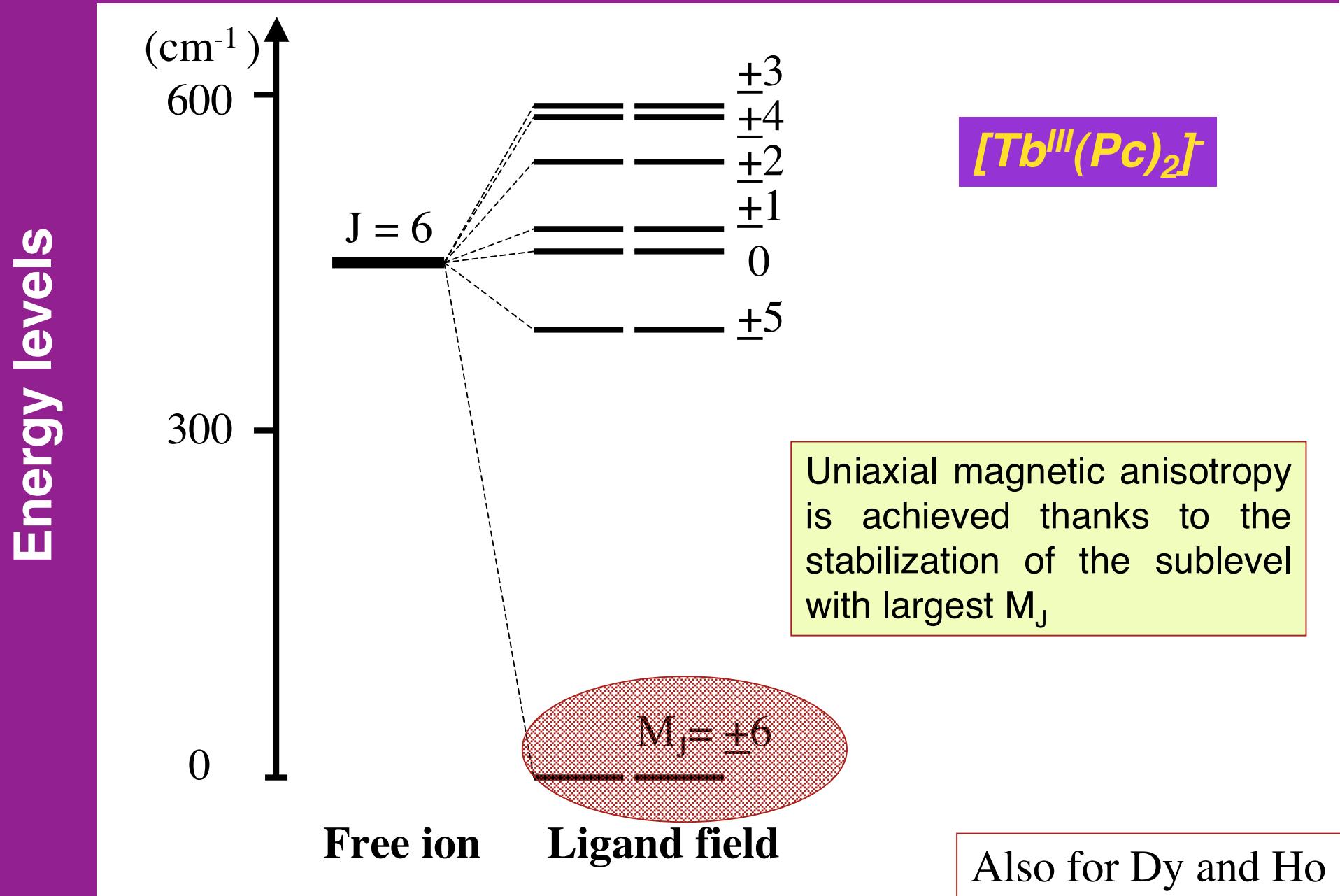
Lanthanide bis-phtalocyanine complexes



D_{4d} symmetry

Koike et. al *Inorg. Chem.* 1996, 35, 5798-5804.

Lanthanide bis-phtalocyanine complexes

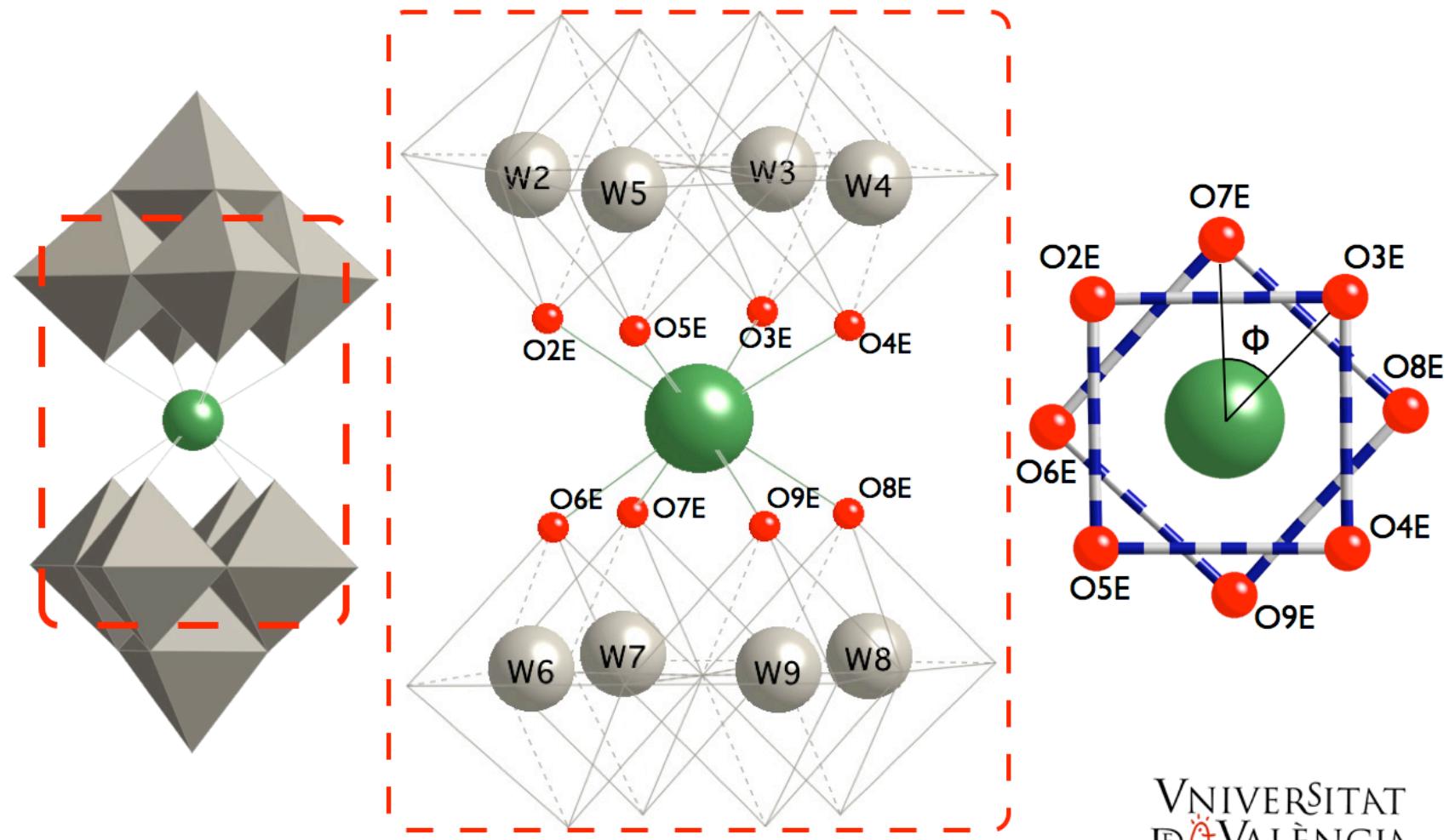


Lanthanide polyoxometalate (POM) complexes

Structural description

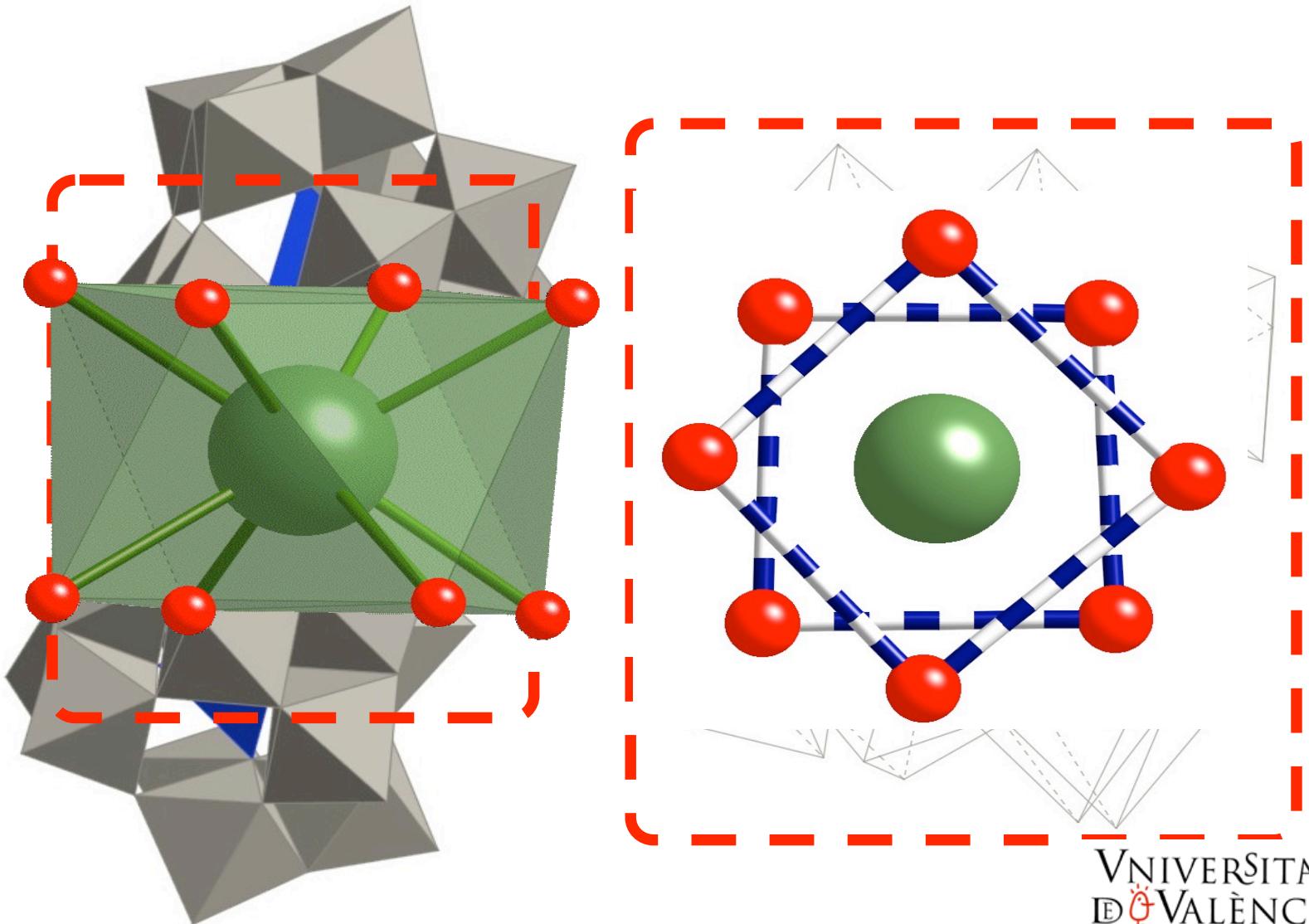
$[Ln^{III}(POM)_2]^{n-}$

$[Ln^{III}(W_{10}O_{36})]Na_9.xH_2O$ ($Ln^{III} = Tb, Dy, Ho, Er$)



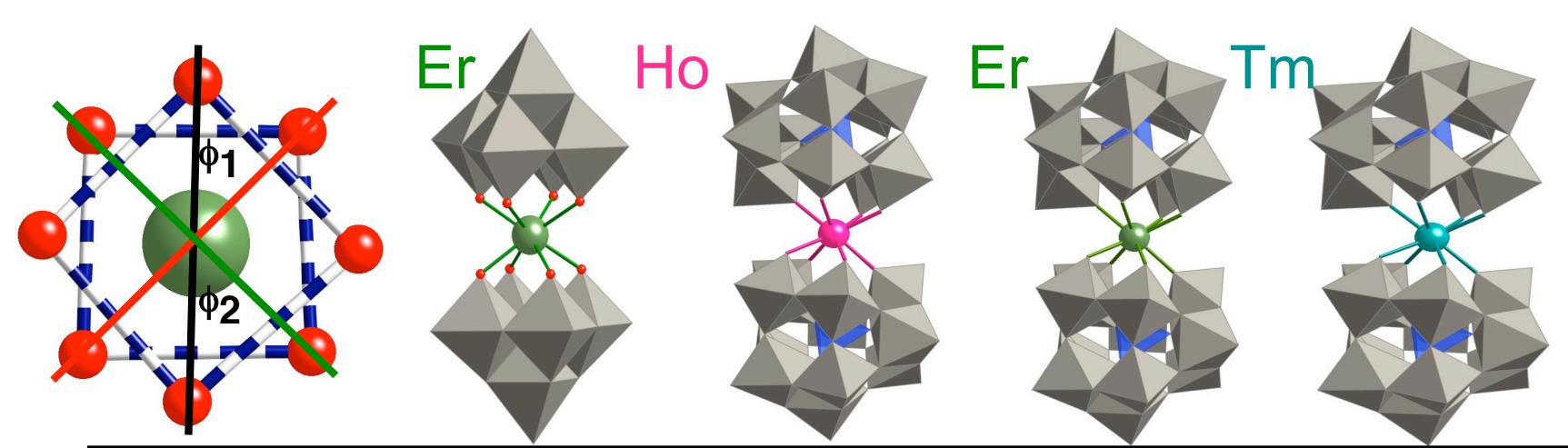
Lanthanide polyoxometalate (POM) complexes

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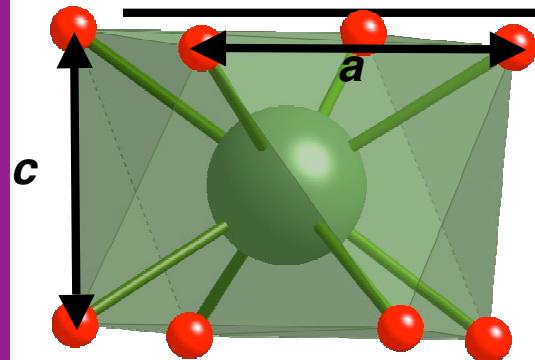


Structural Parameters

$[Ln^{III}(W_{10}O_{36})]^{9-}$ and $[Ln^{III}(SiW_{11}O_{39})_2]^{13-}$ families



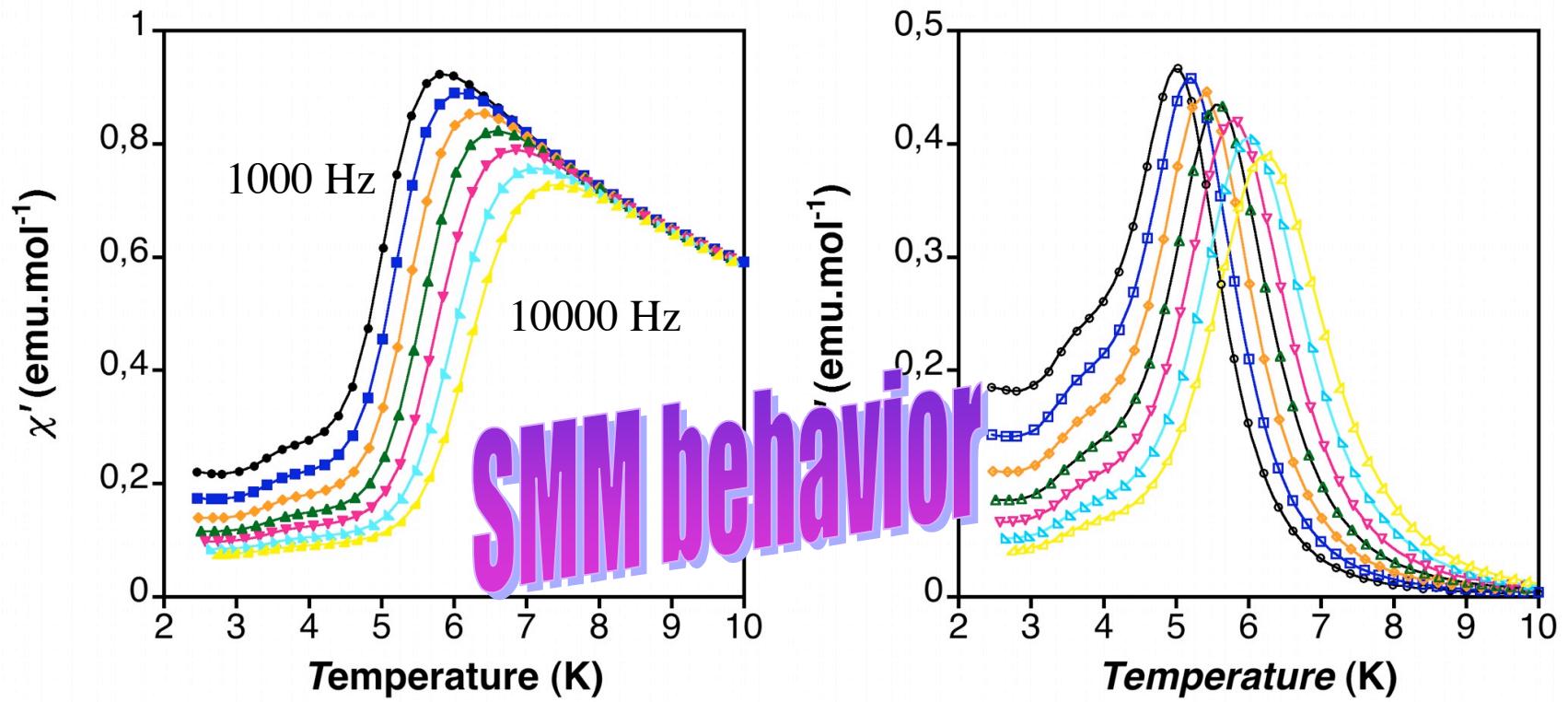
ϕ_1	47,6	46,1	43,1	45,2
ϕ_2	44,2	41,1	45,5	46,0
a (Å)	2,86(5)	2,80(4)	2,82(8)	2,84(3)
c (Å)	2,47(1)	2,39(5)	2,48(5)	2,46(6)



**Compression of the antiprism
(vs. elongation in the phtalocyanine family)**

$[Er(W_{10}O_{36})]Na_9.nH_2O$

Magnetic characterization



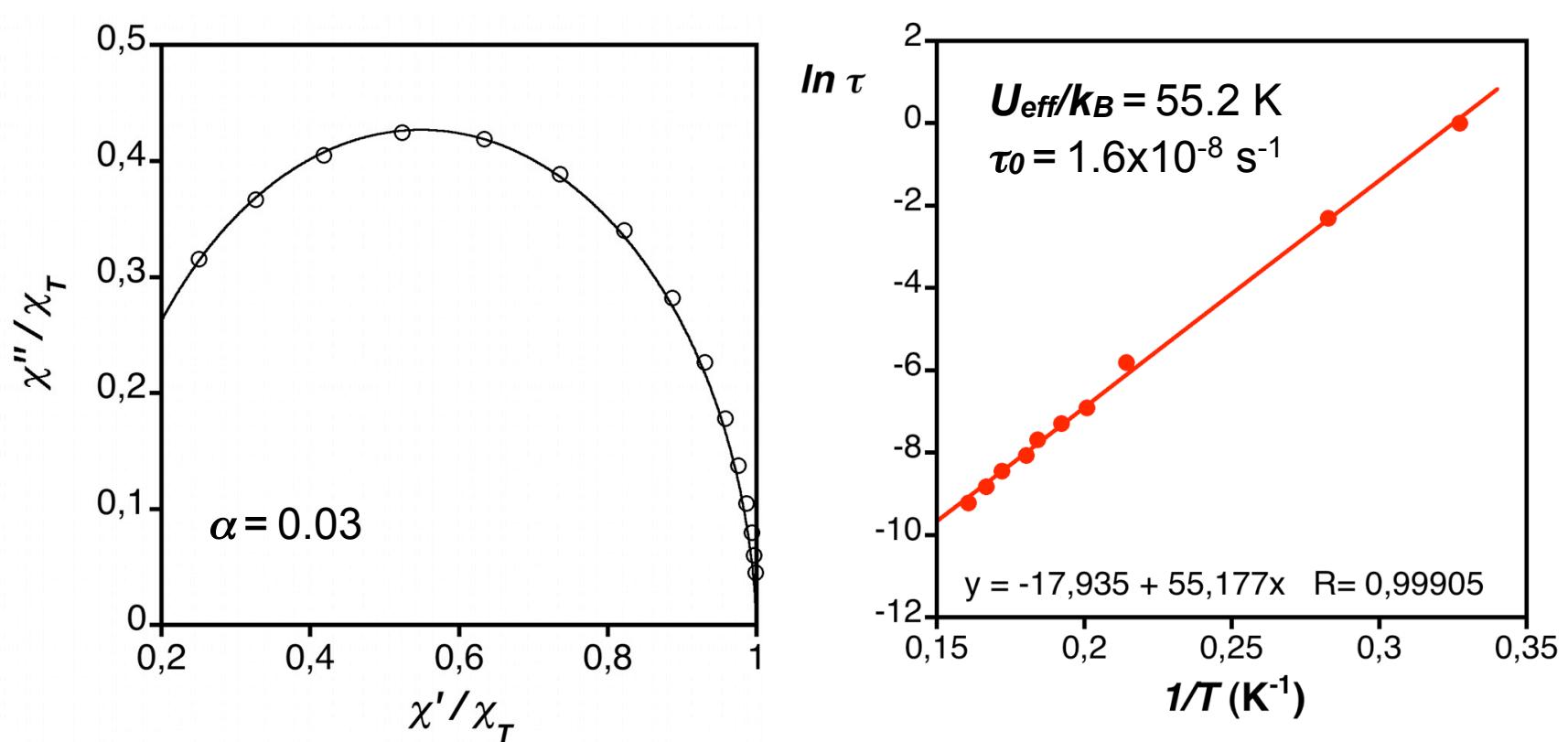
AC magnetic susceptibility at 1000, 1467, 2154, 3162, 4641, 6813 and 10000 Hz

Frequency-dependent superparamagnetic blocking

Slow magnetic relaxation

$[Er(W_{10}O_{36})]Na_9.nH_2O$

Magnetic characterization

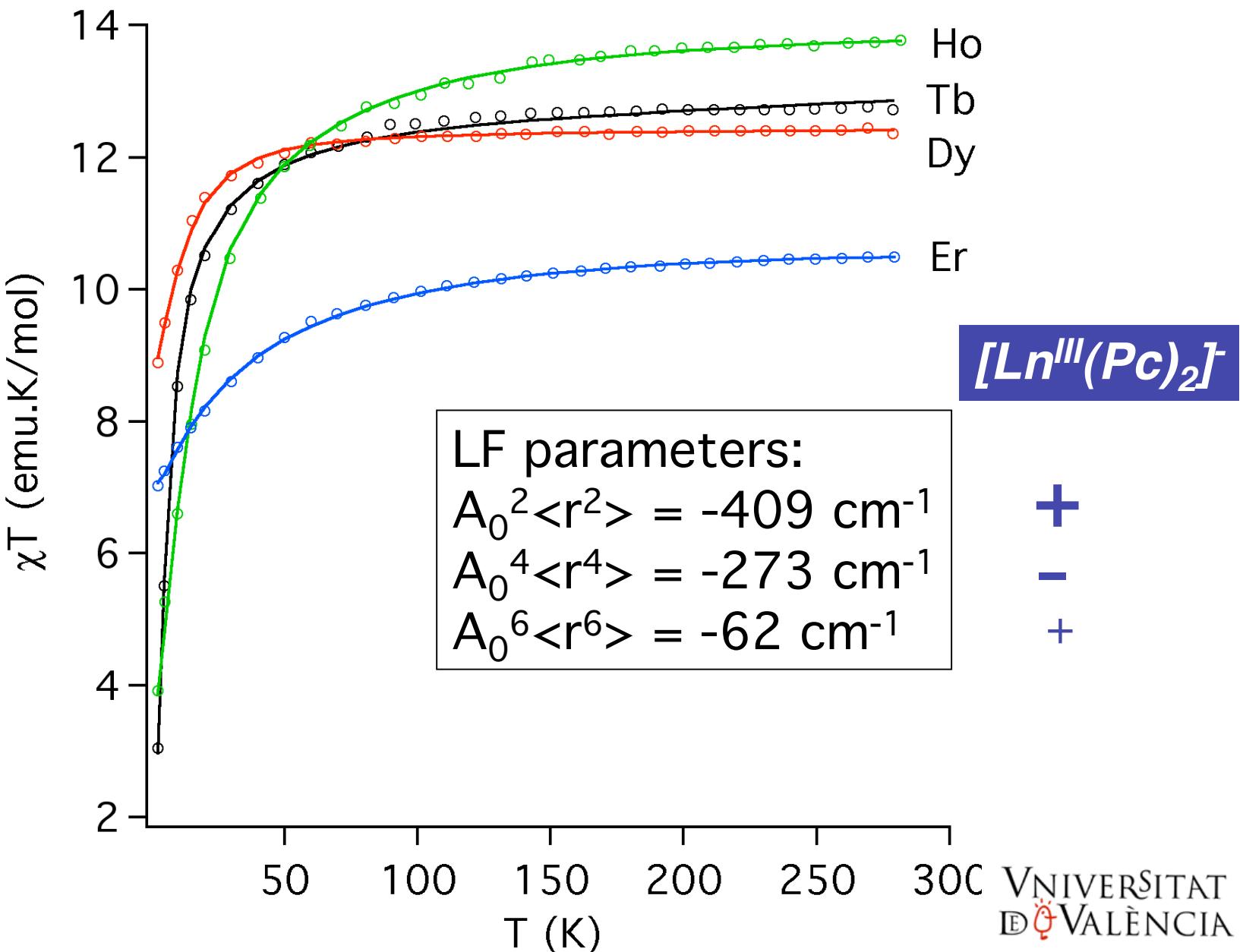


Argand plot at 5 K (*left*). Experimental data best fitting to Arrhenius law for the frequencies 1-10000 Hz (*right*).

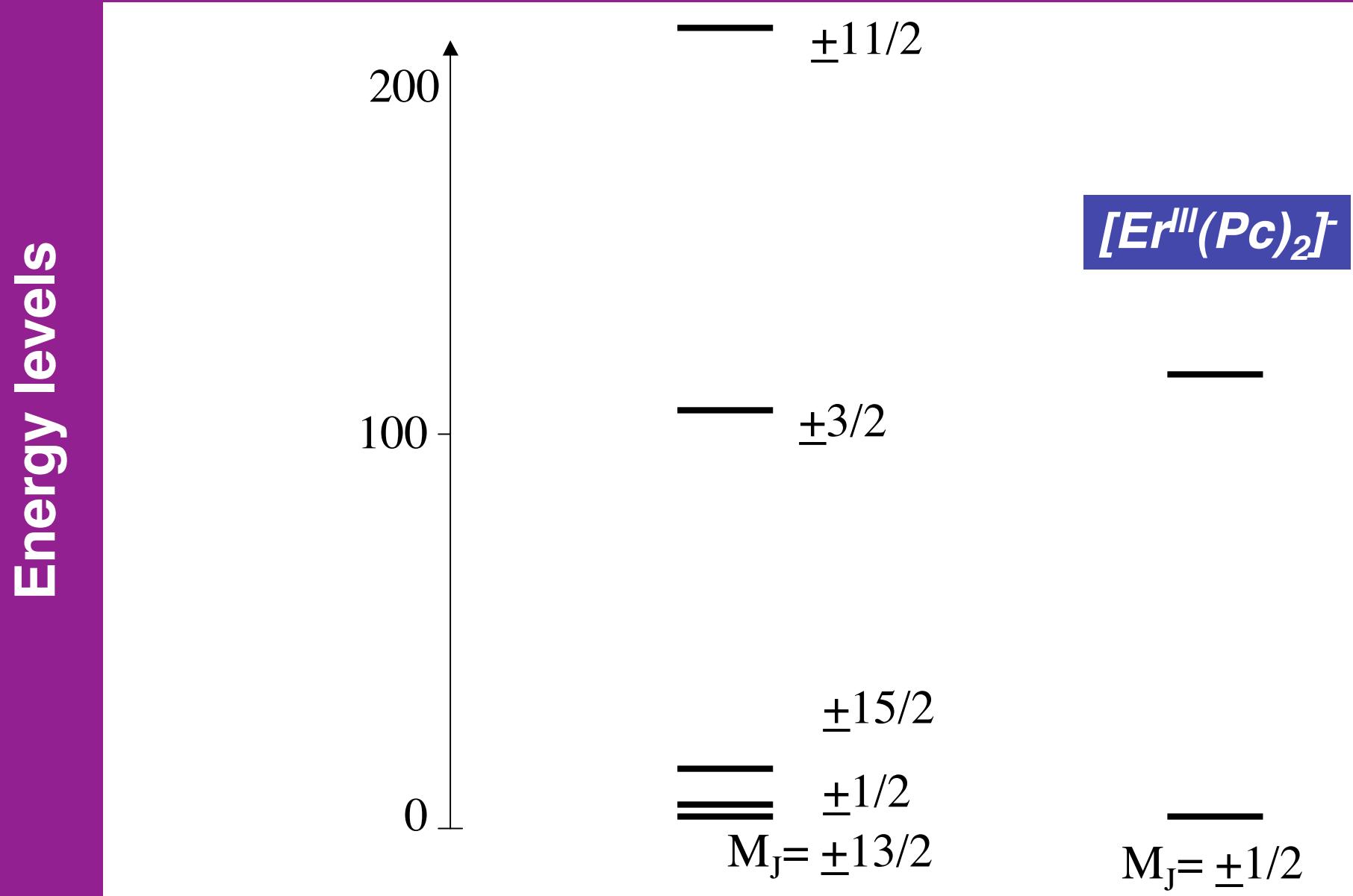
E. Coronado et al. *JACS* **2008**, *130*, 8874

$[Ln(W_{10}O_{36})]Na_9.nH_2O$

Ligand field parameters

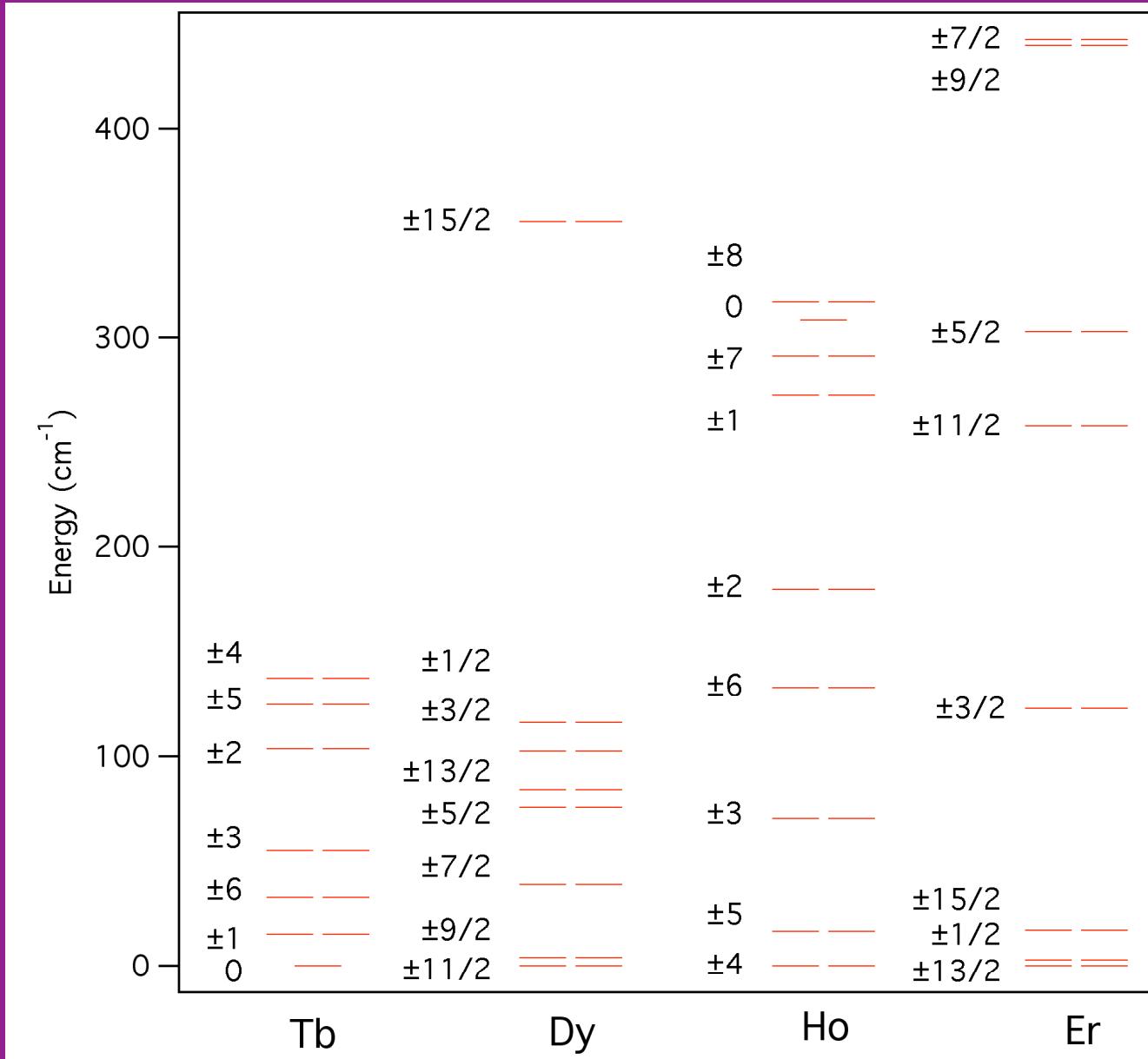


[Er(W₁₀O₃₆)Na₉.nH₂O



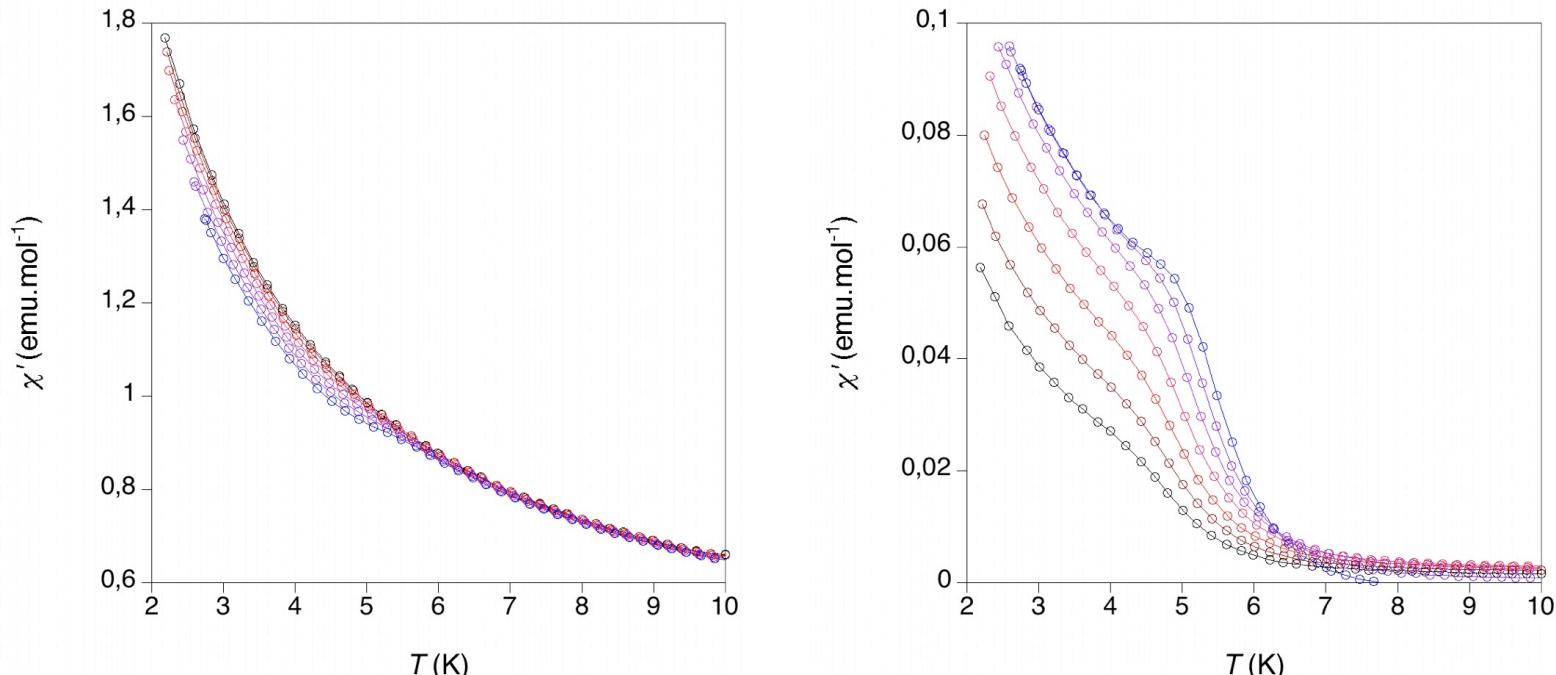
$[Ln(W_{10}O_{36})]Na_9.nH_2O$

Energy levels



Magnetic characterization

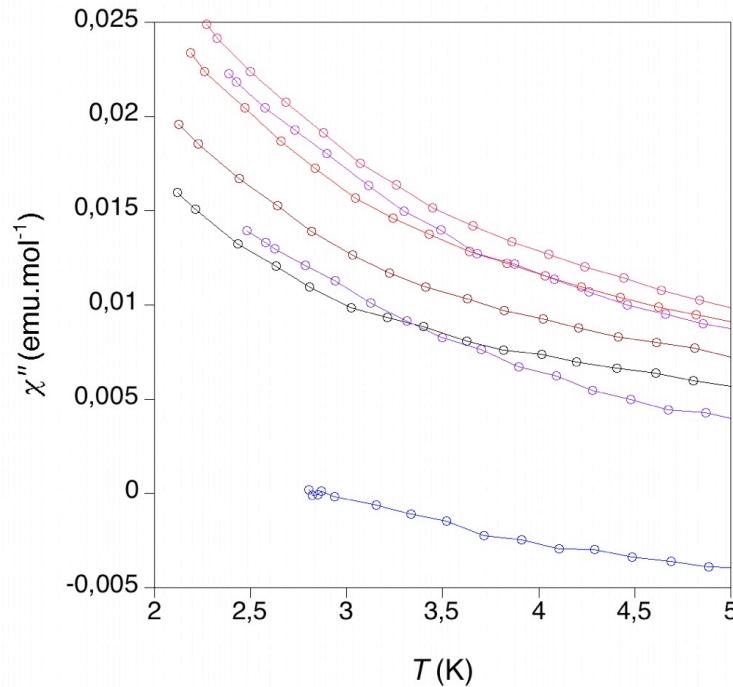
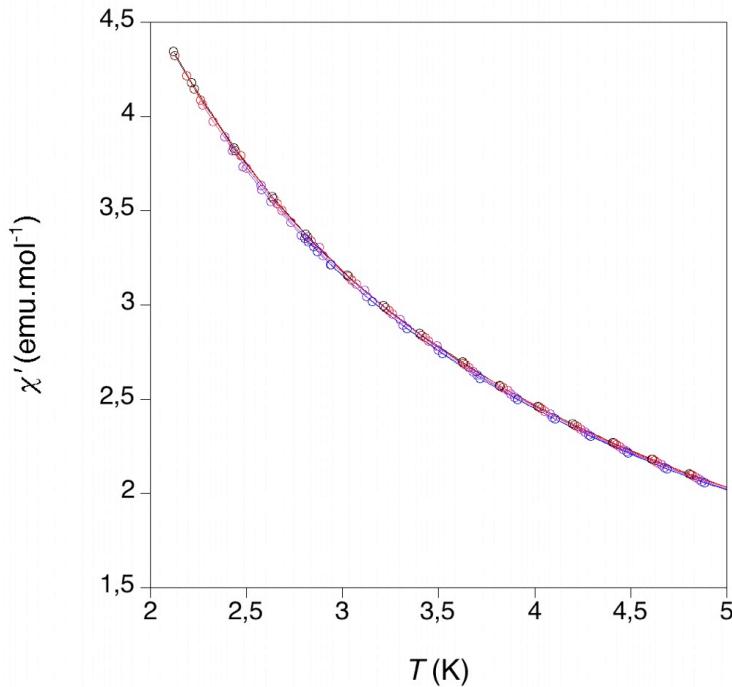
$[Ho(W_{10}O_{36})]Na_9.nH_2O$



AC magnetic susceptibility at 1000, 1467, 2154, 3162, 4641, 6813 and 10000 Hz

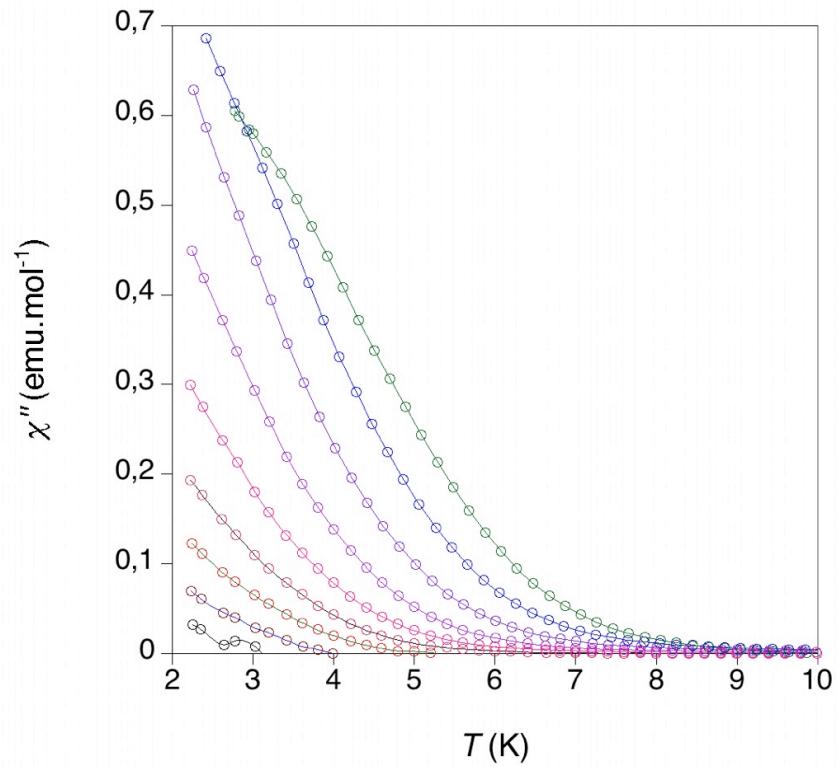
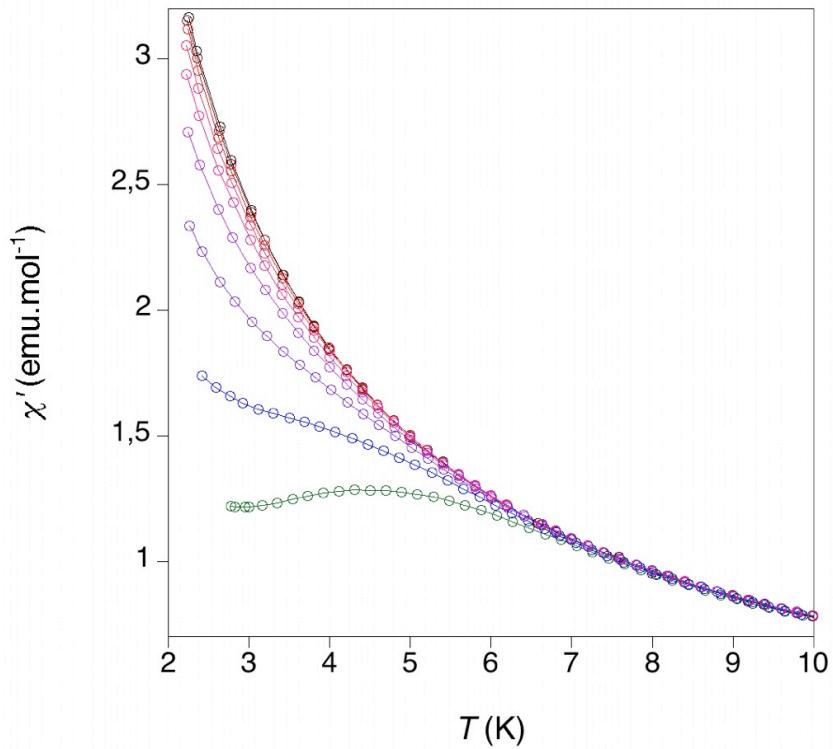
$[Dy(W_{10}O_{36})]Na_9.nH_2O$

Magnetic characterization



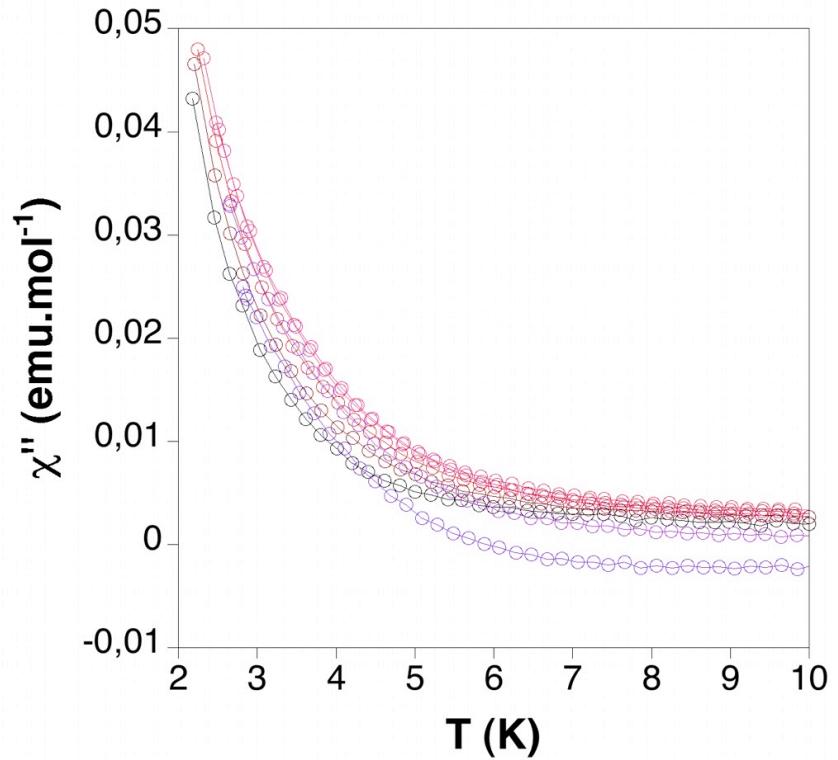
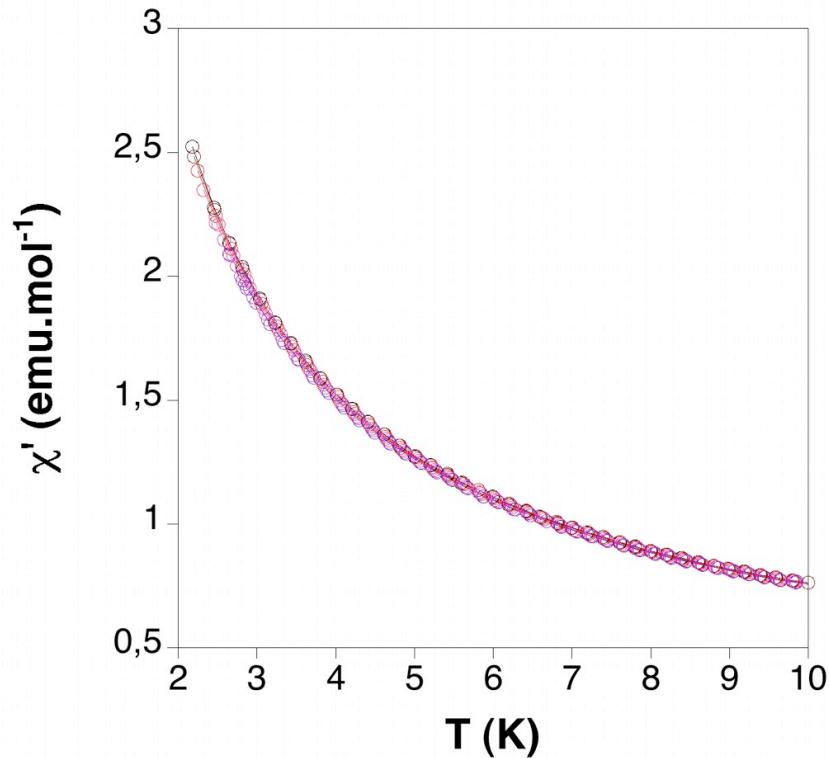
AC magnetic susceptibility at 1000, 1467, 2154, 3162, 4641, 6813 and 10000 Hz

Magnetic characterization



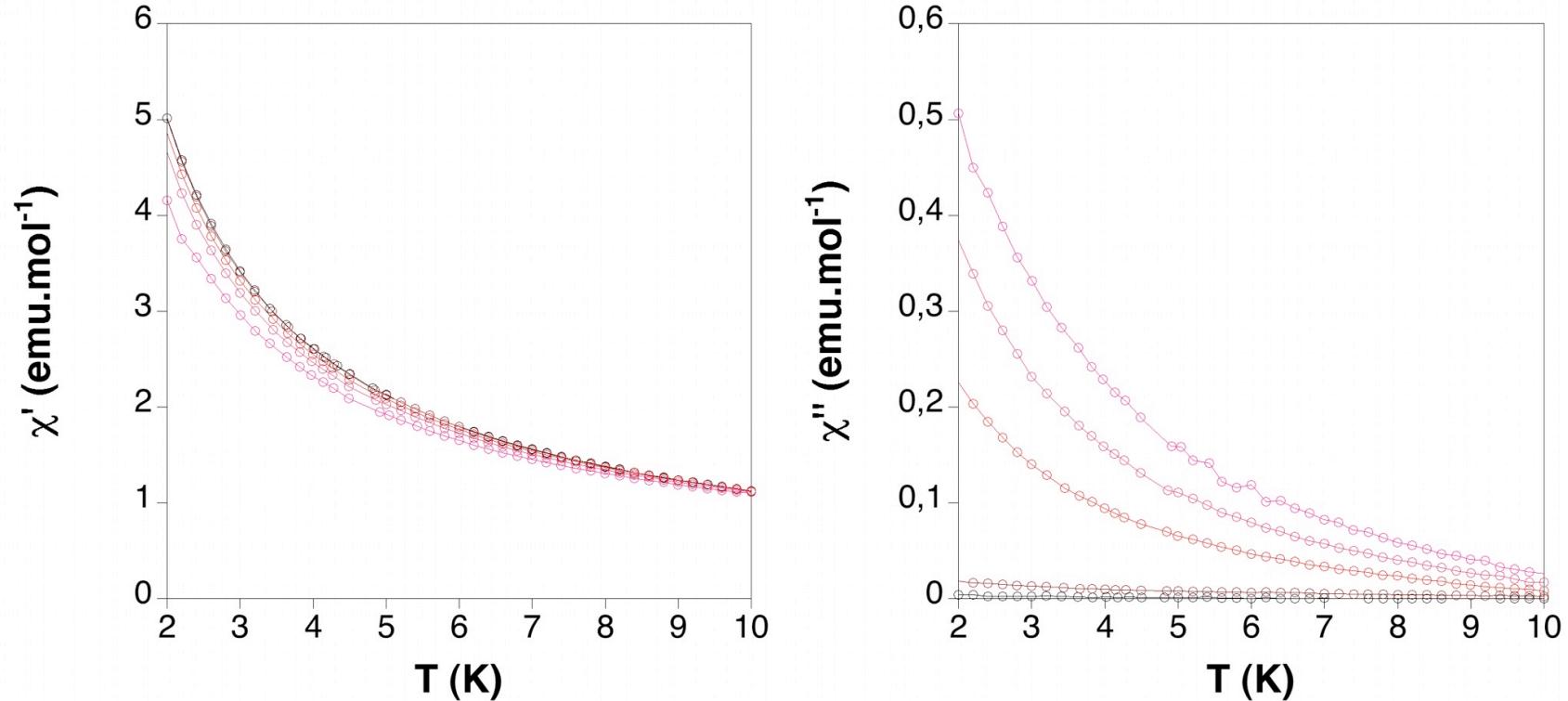
AC magnetic susceptibility at 1000, 1467, 2154, 3162, 4641, 6813 and 10000 Hz

Magnetic characterization



$[Dy(SiW_{11}O_{39})_2]K_{13}.nH_2O$

Magnetic characterization



AC magnetic susceptibility at 1000, 1467, 2154, 3162, 4641, 6813 and 10000 Hz

Outline

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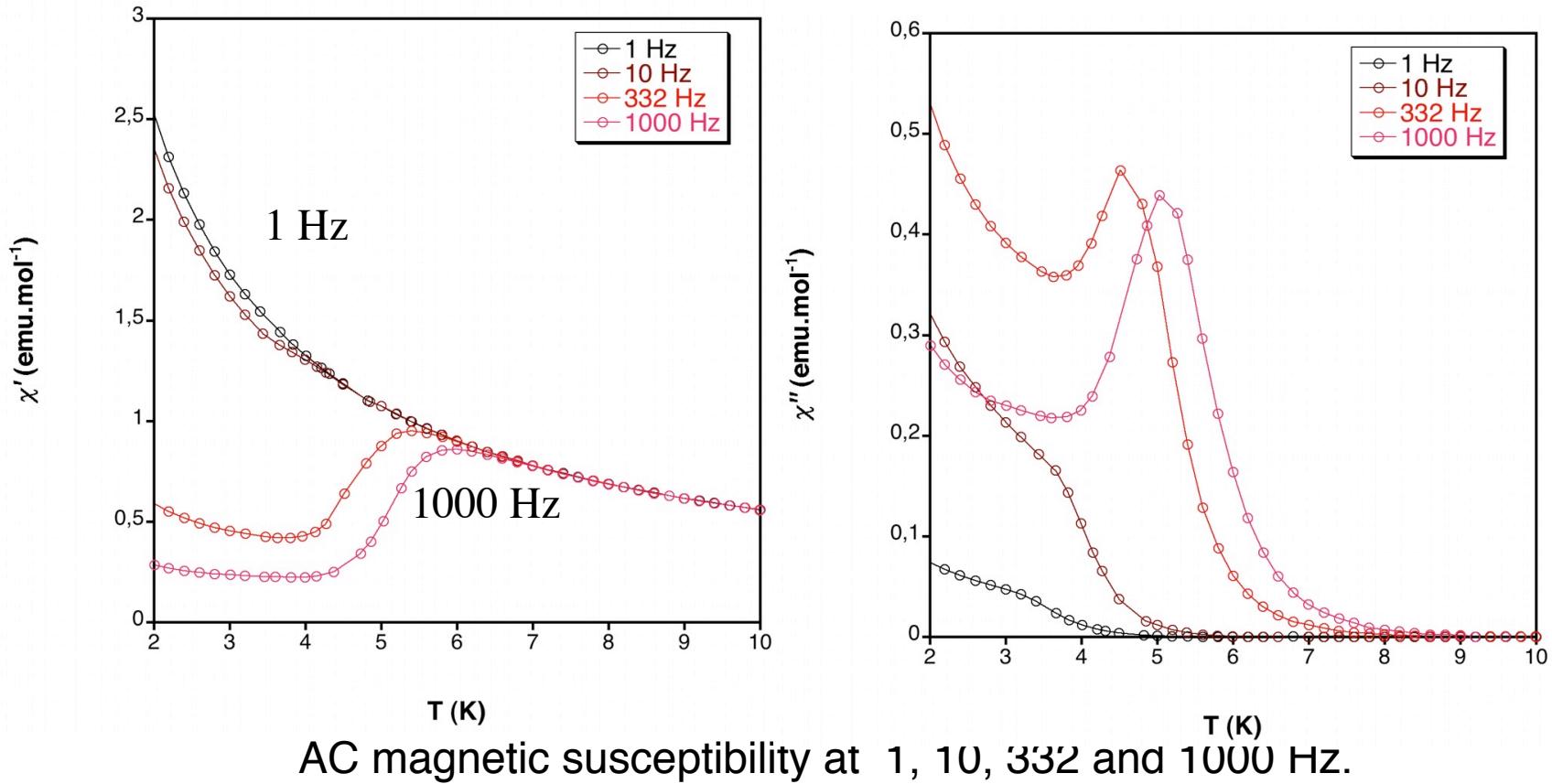
II. SMMs and **Quantum tunnelling**
in lanthanide POMs

collaboration with Fernando Luis

III. SMMs into superconductors

Exotic magnetic behavior

$[Er(W_{10}O_{36})]Na_9.nH_2O$



Divergence in χ' and χ'' below the blocking T

Fast magnetic relaxation at low frequencies



Divergence in χ' and χ'' below the blocking T

**ORIGIN: VERY FAST QUANTUM TUNNELLING
OF THE MAGNETIZATION via THE GROUND STATE**

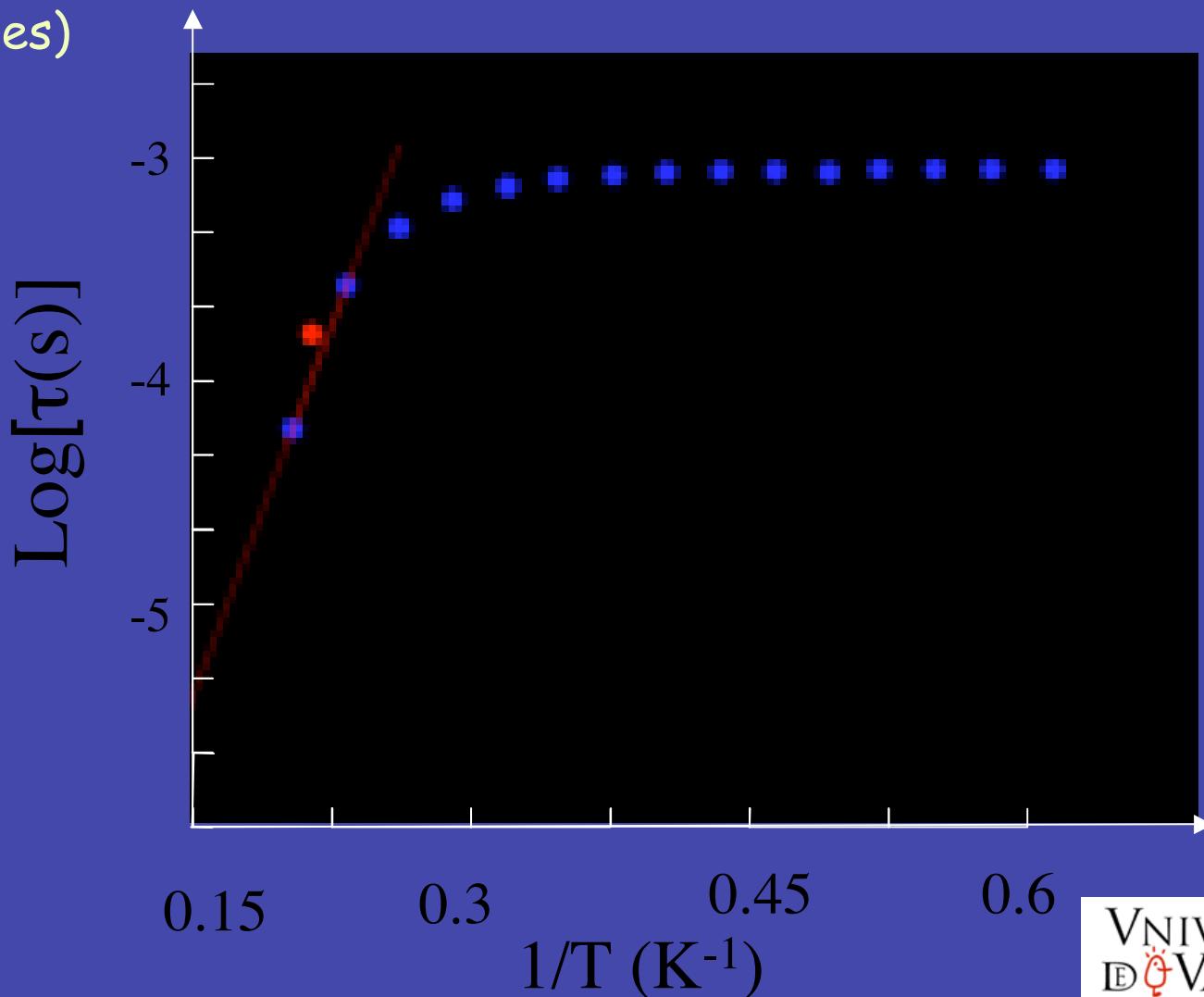
**Susceptibility fully blocked only when
the time scale of the experiment
 $(1/2\pi\nu) \ll \text{tunnelling time}$
(at $\nu > 10000 \text{ Hz}$)**

**Susceptibility unblocked (even at $T = 0$) when
 $(1/2\pi\nu) > \text{tunnelling time}$
(at $\nu > 10 \text{ Hz}$)**

[Er(W₁₀O₃₆)Na₉.nH₂O

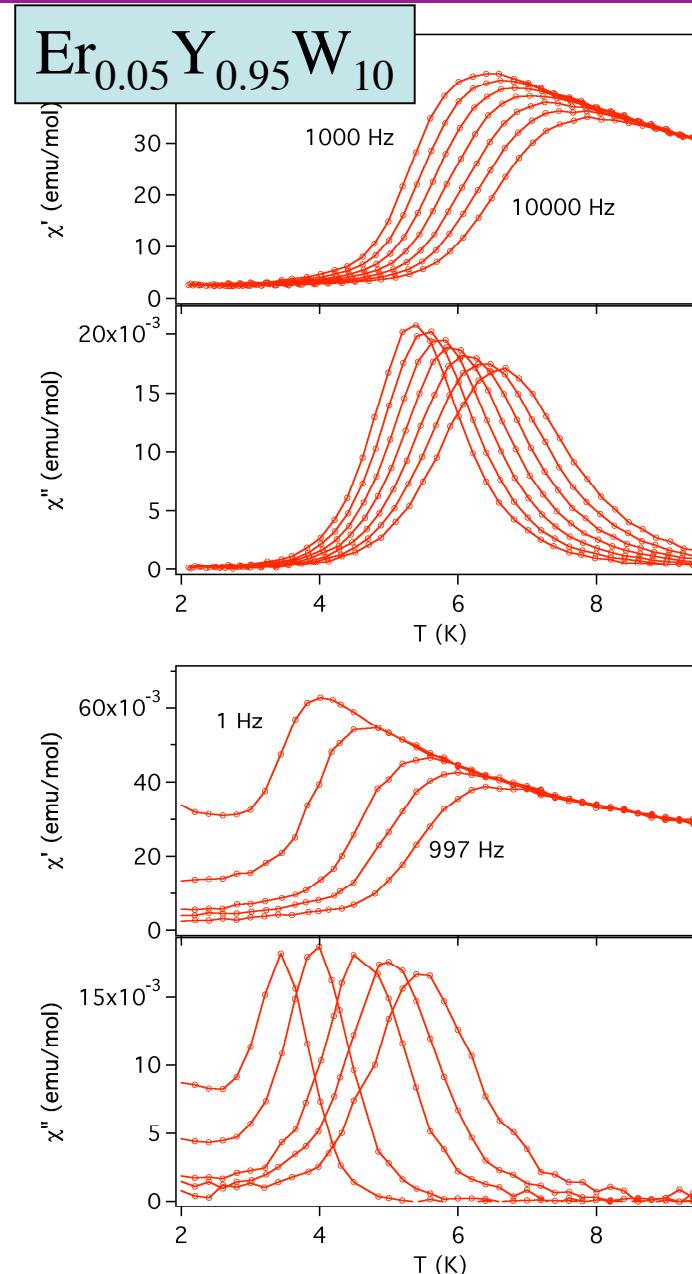
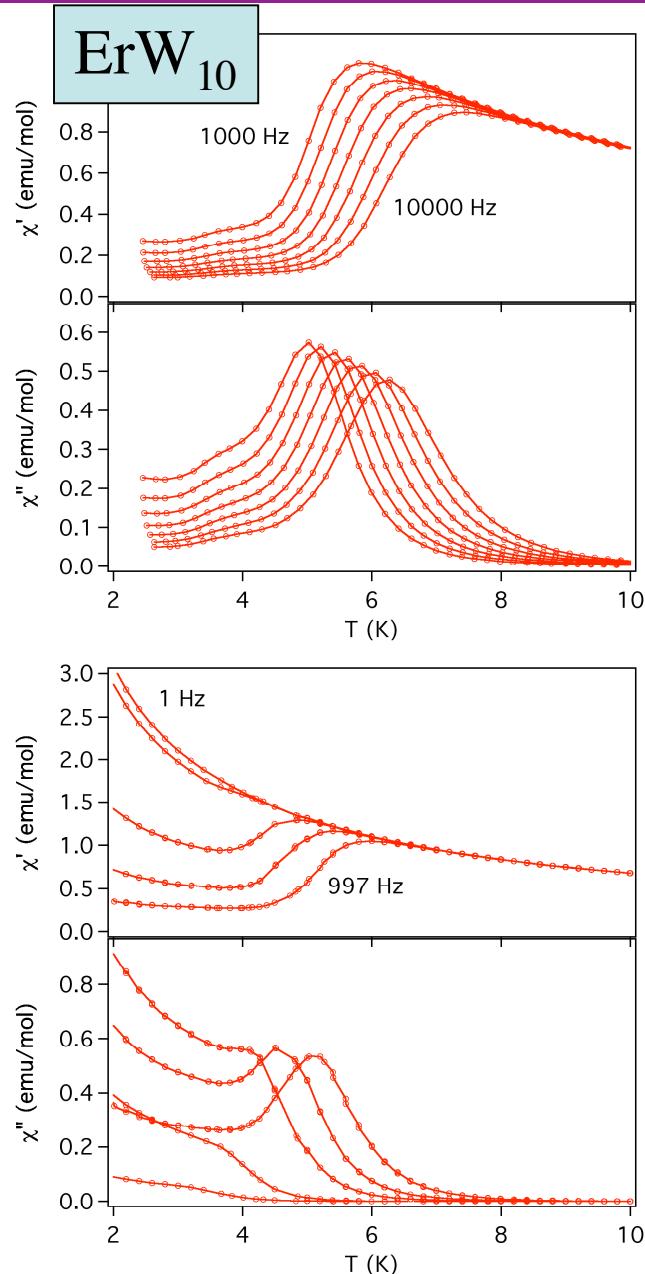
magnetic behavior

Quantum tunnelling regime below 4.4 K with a relaxation time of *ca* 1 ms. (from the study of the ac susceptibilities as a function of frequencies)



MAGNETIC DIPOLAR EFFECTS

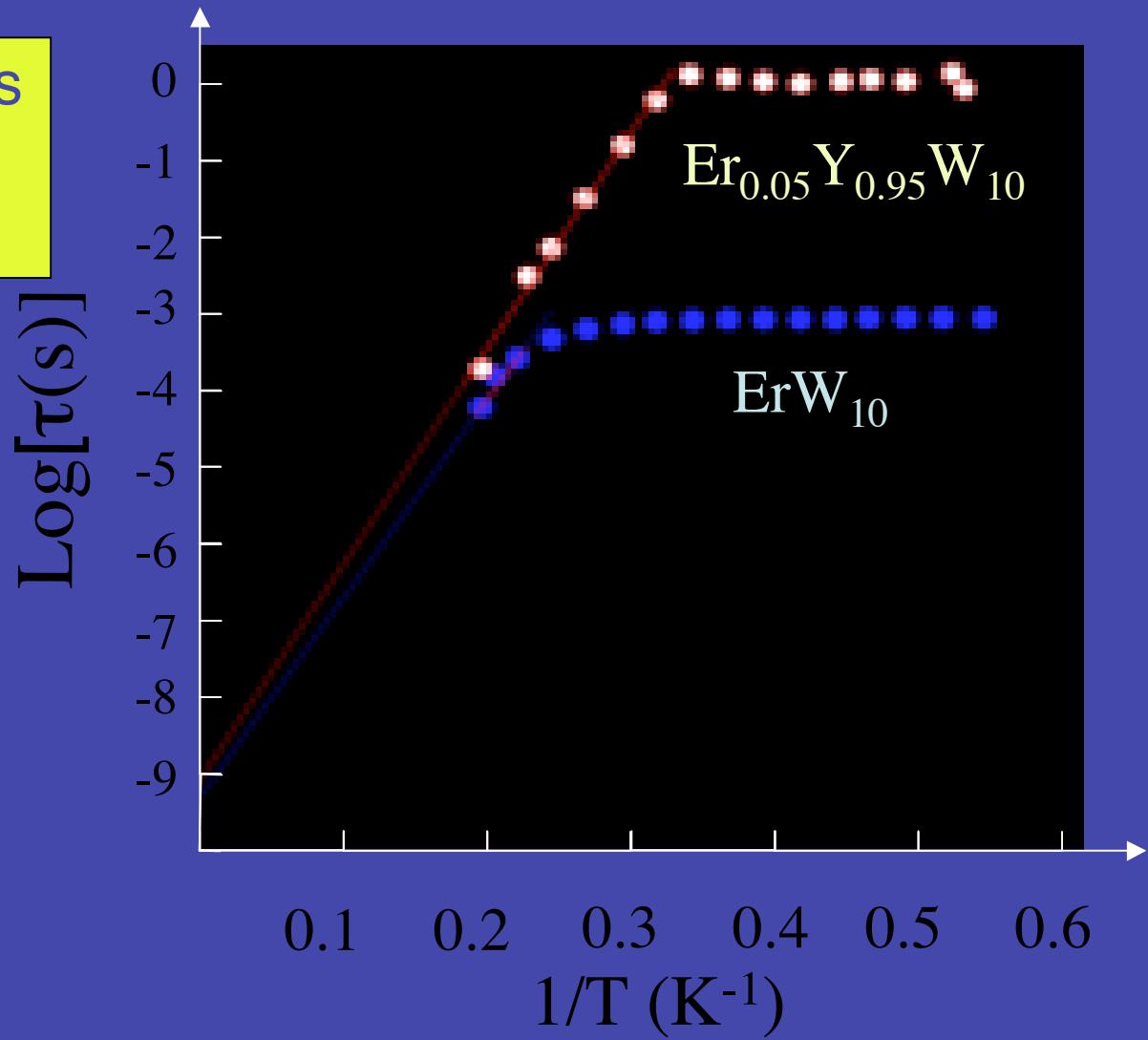
$[Er(W_{10}O_{36})]Na_9.nH_2O$



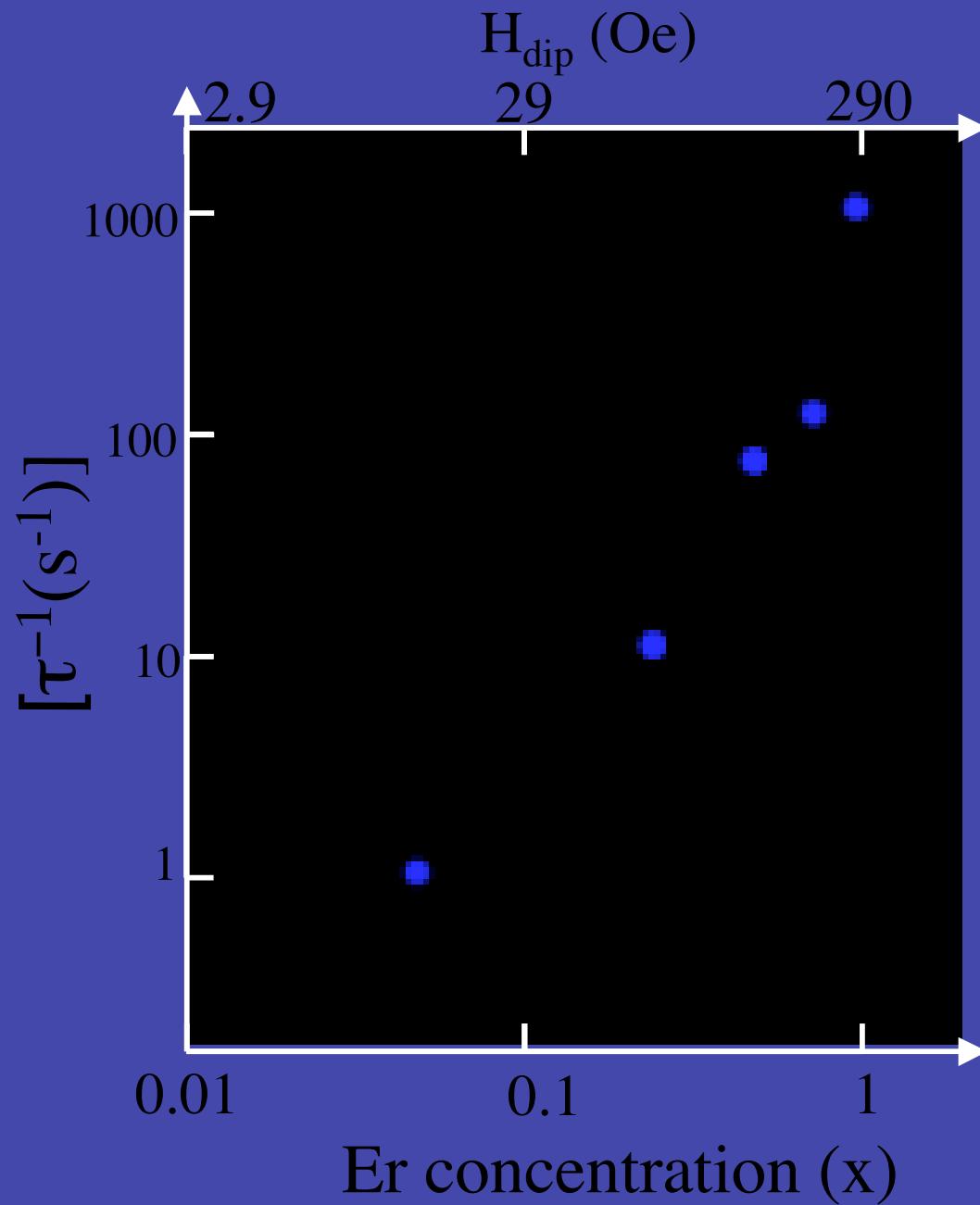
MAGNETIC DIPOLEAR EFFECTS

$[Er(W_{10}O_{36})]Na_9.nH_2O$

Dipole interactions open a “window” for the tunnelling

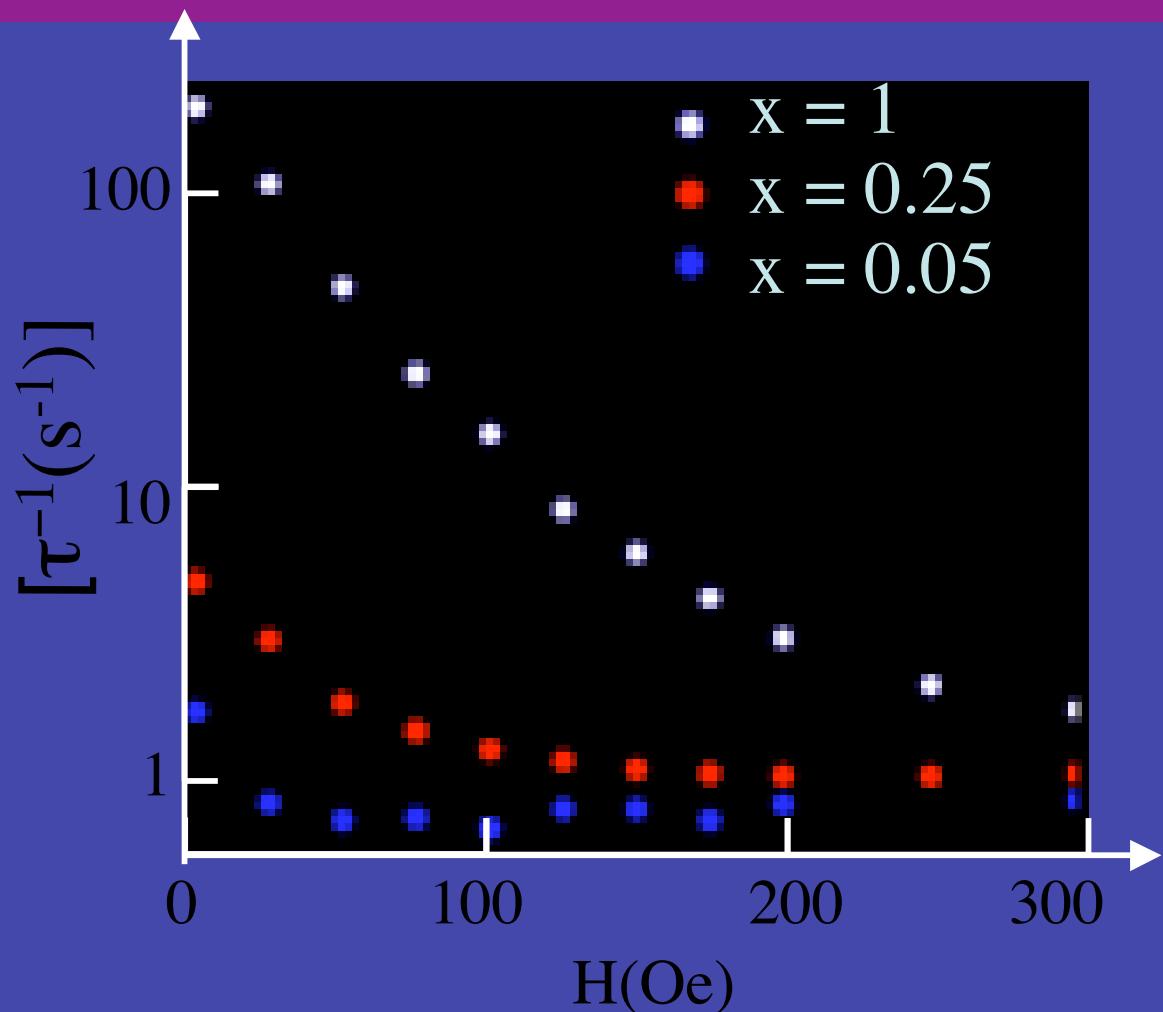


MAGNETIC DIPOLEAR EFFECTS



The spin-lattice relaxation rate is largely enhanced by dipole-dipole interactions acting as an “effective” magnetic field

EXTERNAL MAGNETIC FIELD EFFECTS



The spin-lattice relaxation rate is rapidly suppressed by application of external magnetic fields

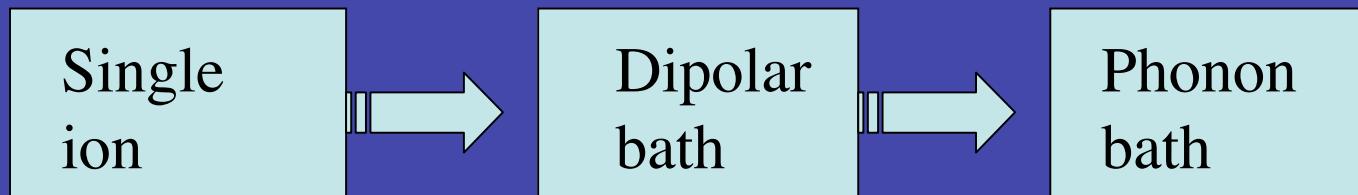
CONCLUSION



The relaxation in the quantum tunnelling regime is very sensitive to extremely weak environmental magnetic fields (**dipolar and external**)

However these two fields produce opposite effects in the spin-lattice relaxation.

MODEL: Dipolar-mediated quantum spin-lattice relaxation.
For sufficiently low magnetic fields spins transfer their energy to the “dipolar bath” via quantum tunnelling processes

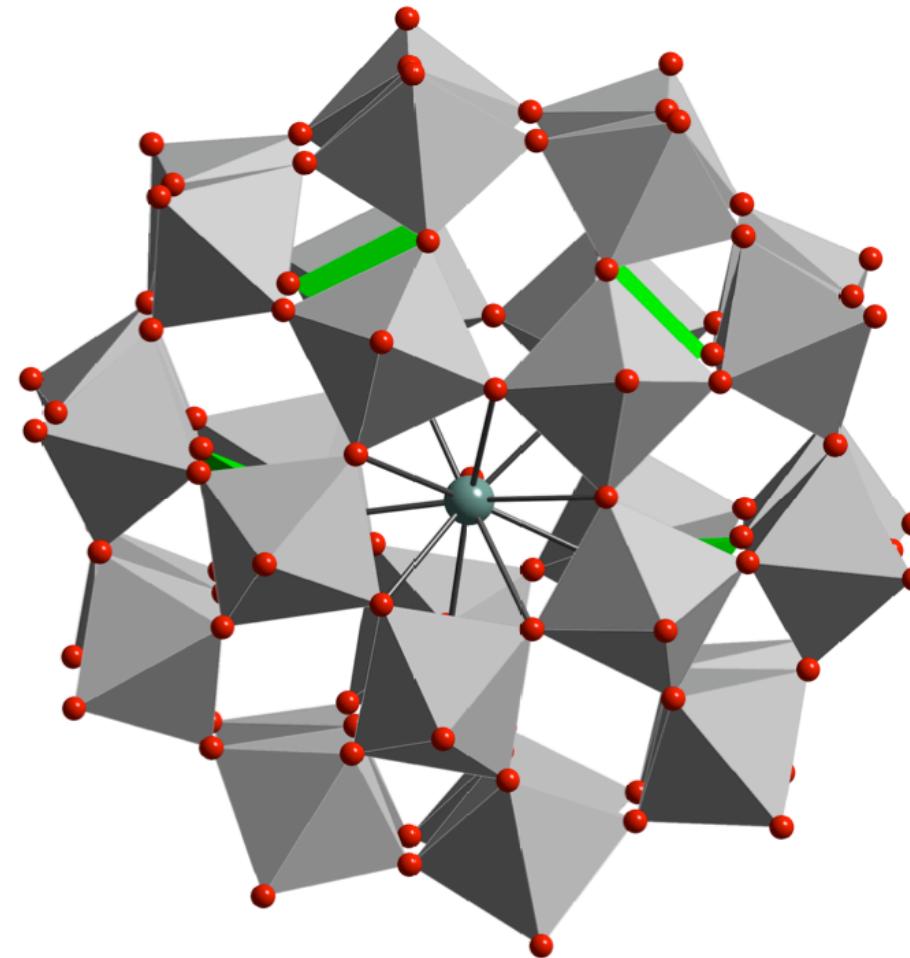


Lanthanide polyoxometalate (POM) complexes

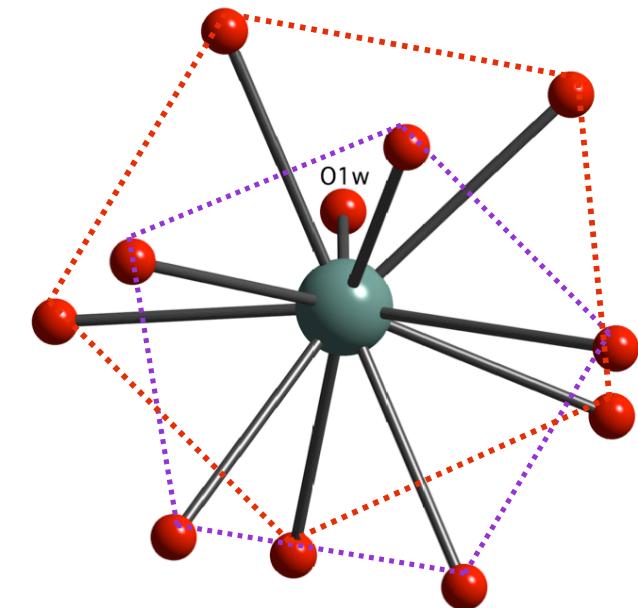
BIGGER CLUSTERS; OTHER CRYSTAL FIELD SYMMETRIES

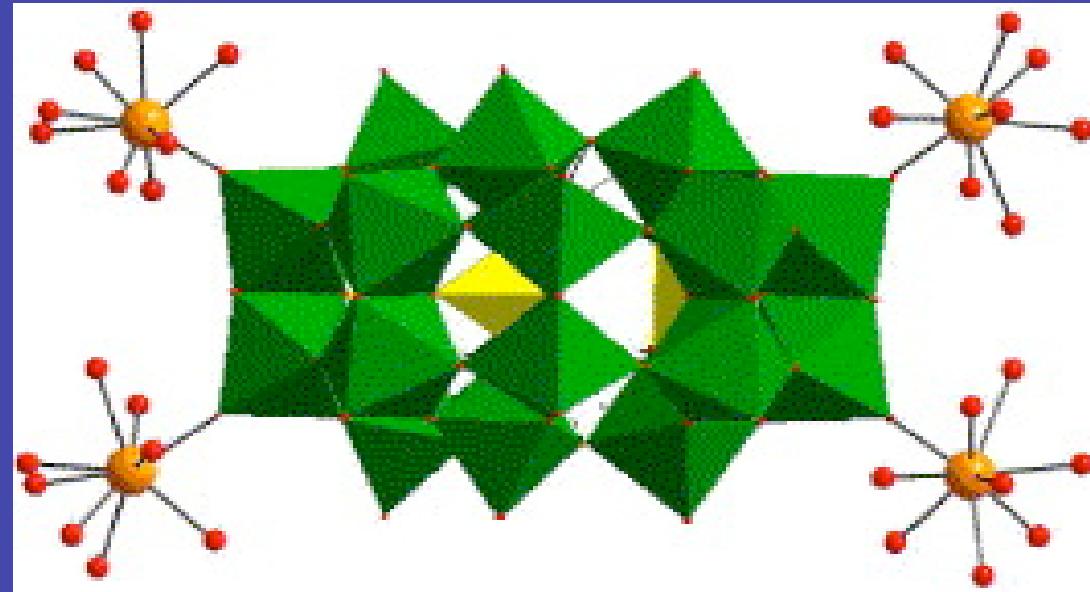
Structural description

Preyssler anion



$[Ln(P_5W_{30}O_{110})]^{n-}$





Adding magnetic moments around SMMs : Single Lanthanide clusters surrounded by magnetic ions (Gd)

POTENTIAL OF LANTHANIDE-POMs

- Extremely good insulation of the rare earth (dilution is not needed).
Even the pure magnetic sample is diluted. Possibility to study spin dynamics in crystal lattices avoiding structural disorder
- Stability of the POM in solution and in the solid state (and in surfaces)
- Spin nuclear-free samples available
- Large single crystals available
- Highly charged anions : Electrospray techniques /
Electrostatic approaches may be used for deposition on surfaces

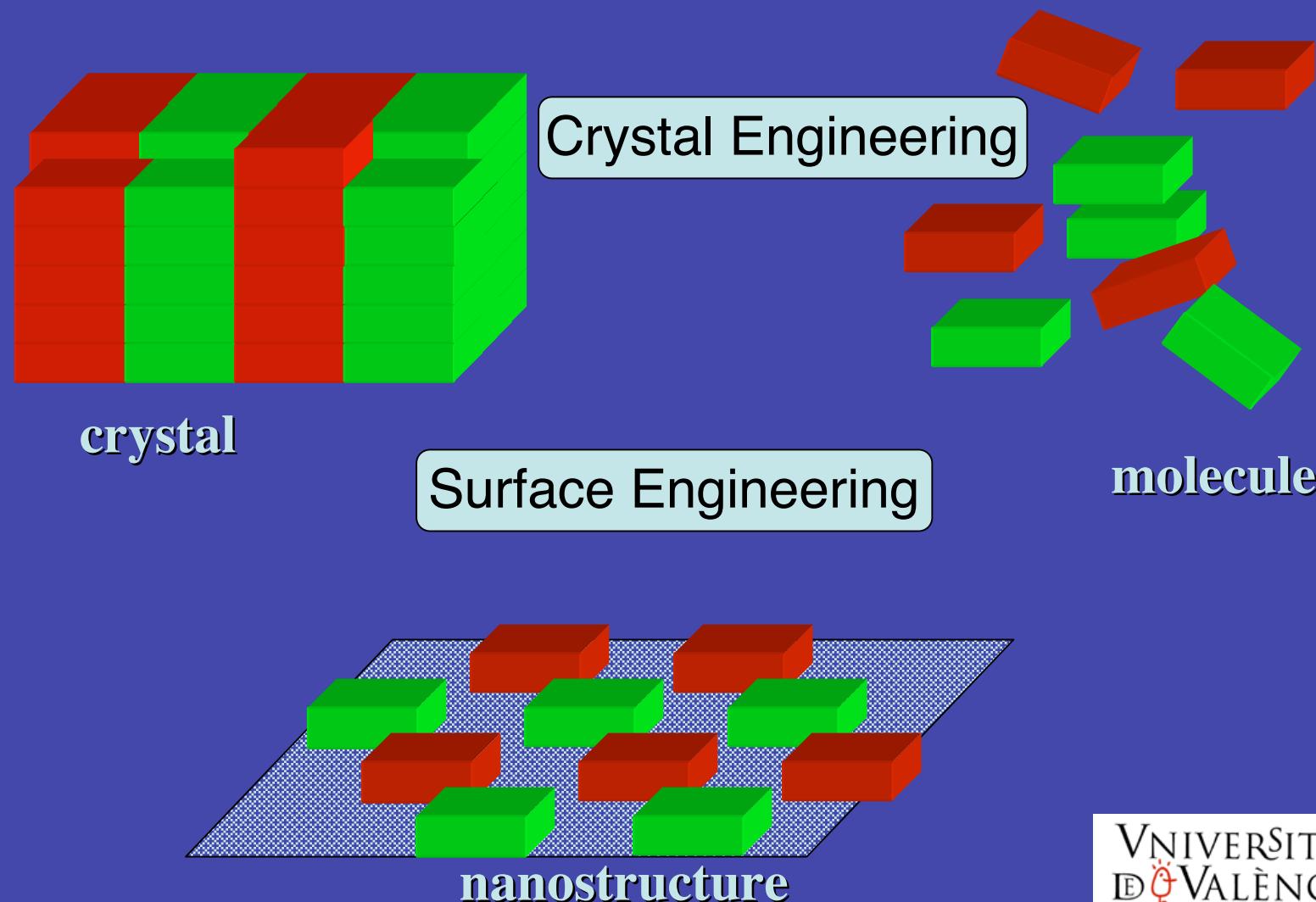
**Ideal systems for studying quantum spin dynamics
in SMMs, for applications in quantum computing, for
making chemistry (hybrid materials), for processing
on surfaces,...**

Molecular Nanomagnets

E. Coronado

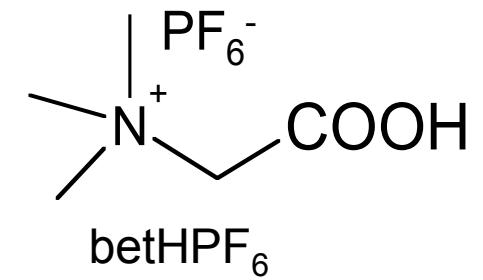
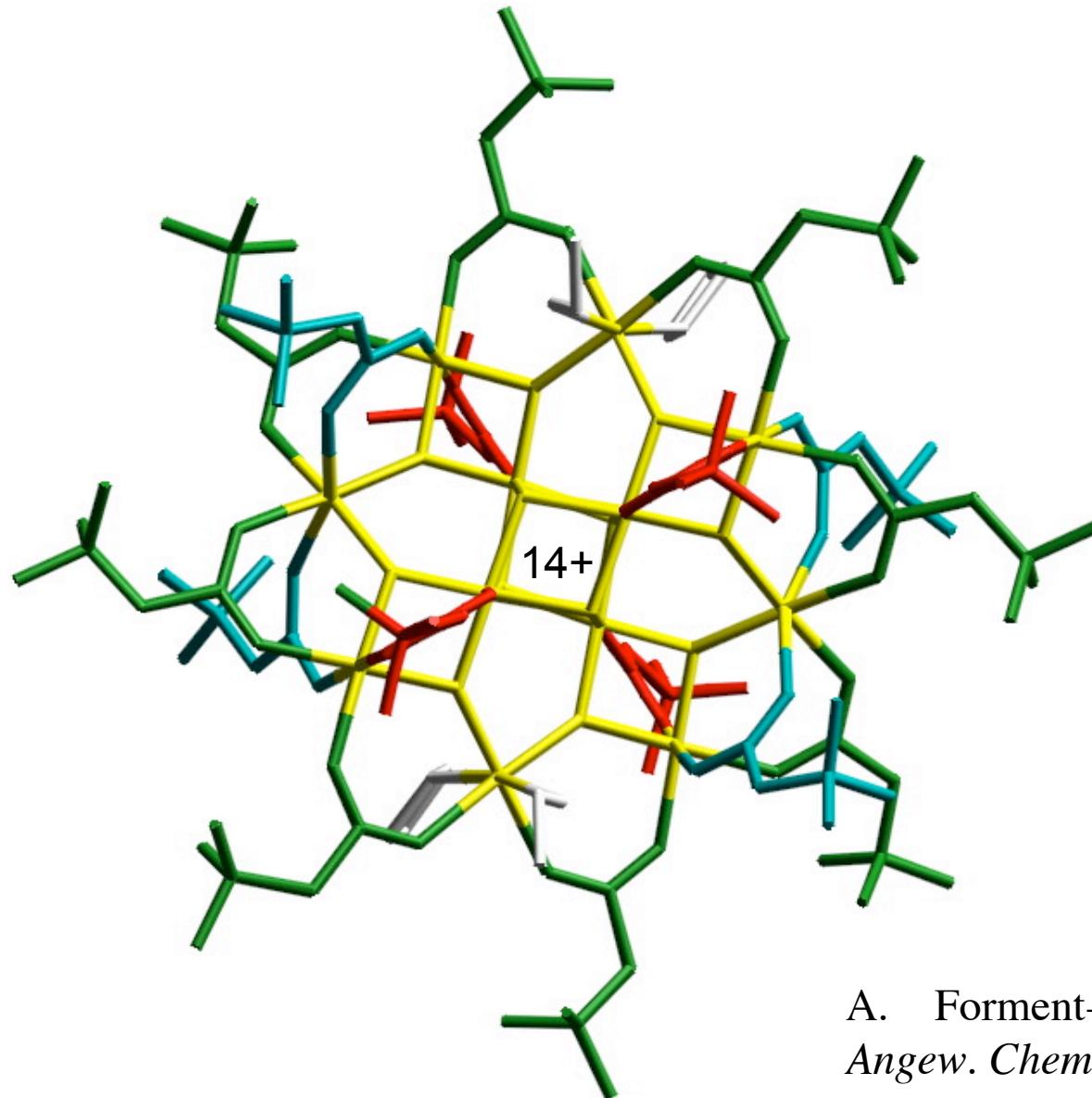
- POLYOXOMETALATE CLUSTERS
- SMMs in different environments
 - On surfaces
 - Intercalated / embedded





Intercalation of Single Molecule Magnets into layered superconductors

Cationic Mn_{12} molecules

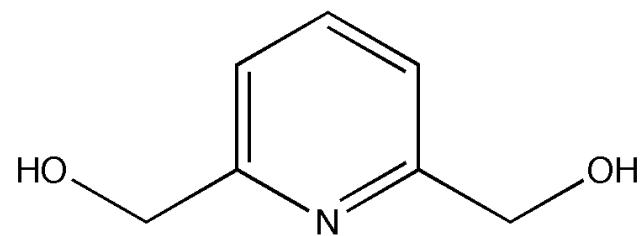
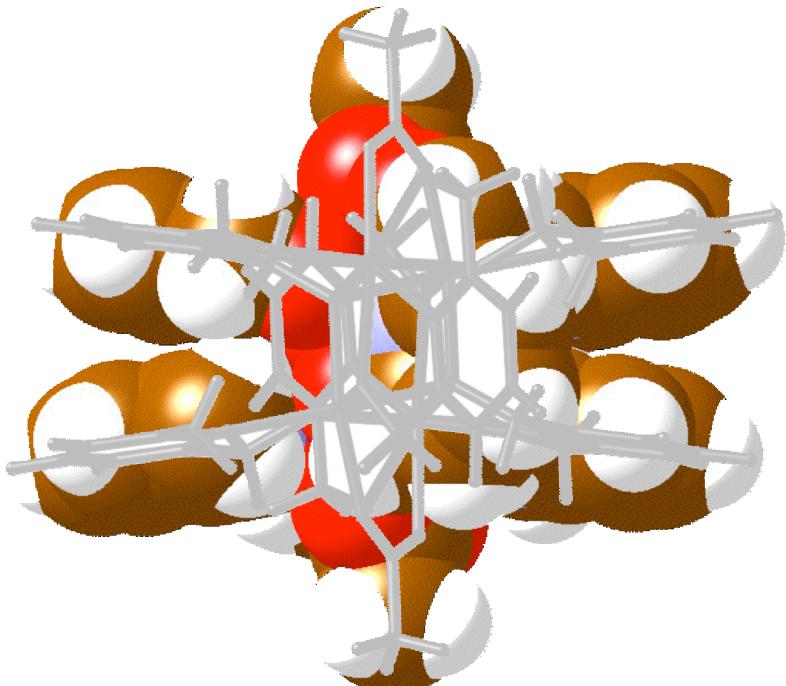
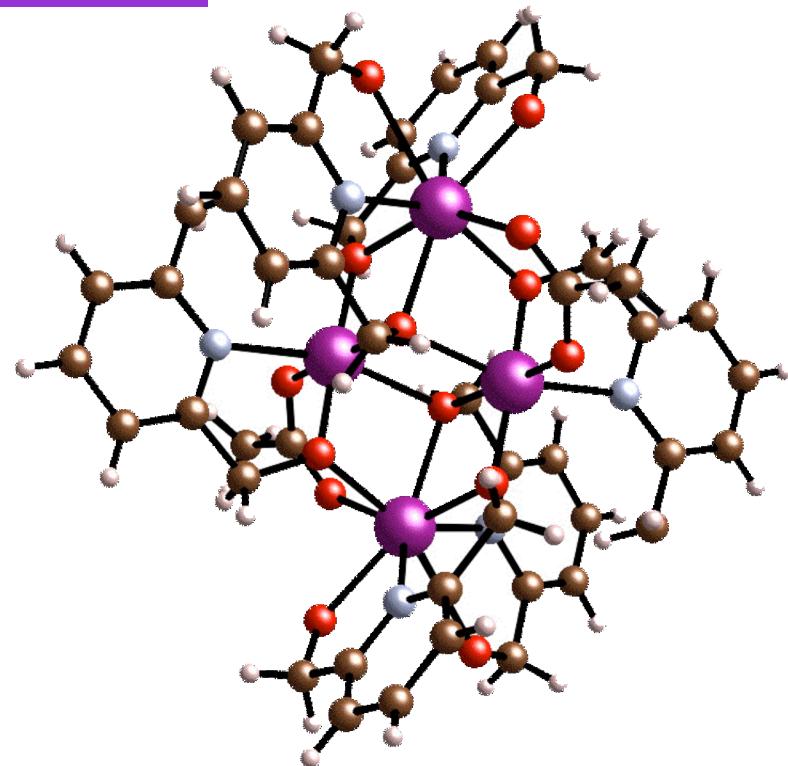


A. Forment-Aliaga et al.
Angew. Chem. Int. Ed. **2004**, *43*, 6152

Structural description

CATIONIC Mn₄ MOLECULES

Mn₄
2+

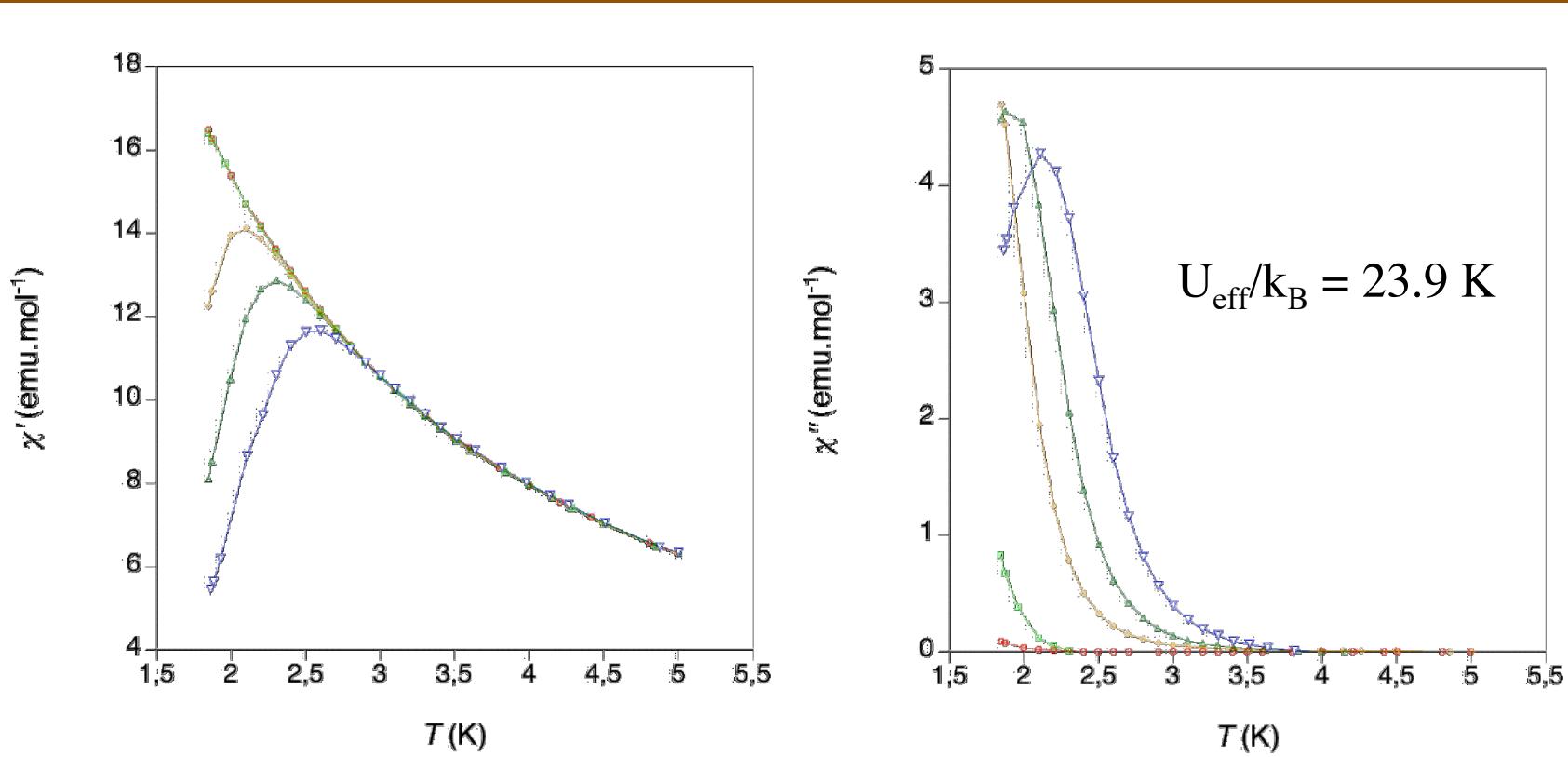


$$\text{pdmH} = \text{C}_7\text{H}_9\text{N}_2$$

*G. Christou; D. N. Hendrickson *et al.* *Inorganic Chemistry* **2000**, 39, (16), 3615-3623.

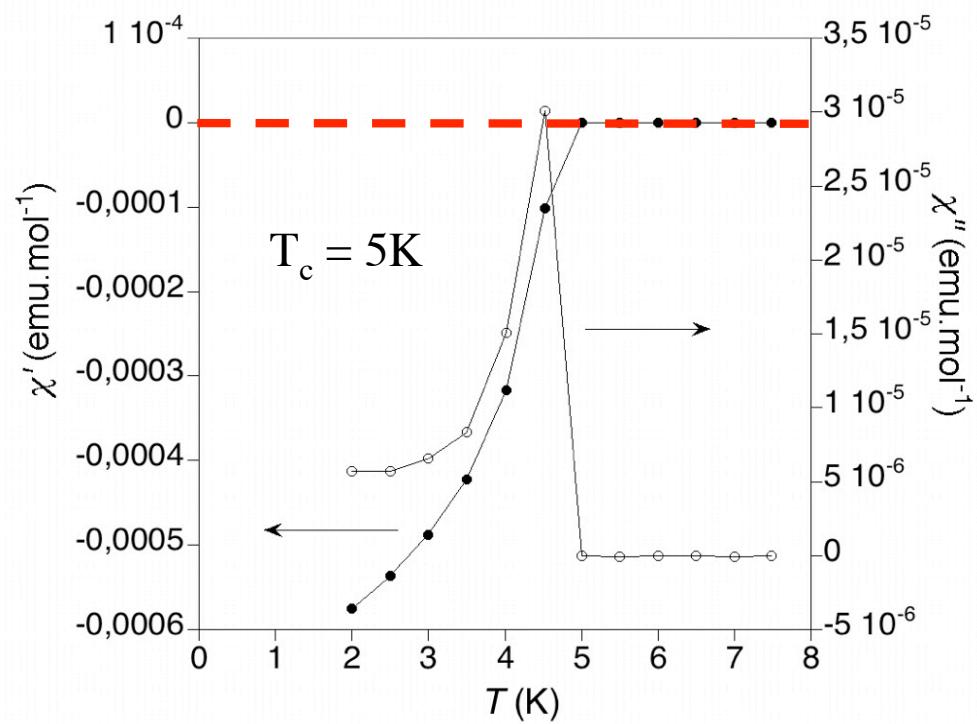
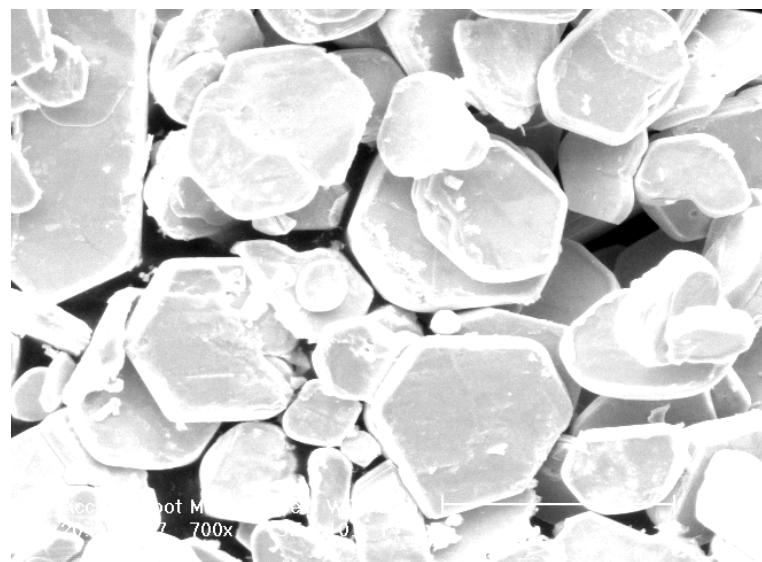
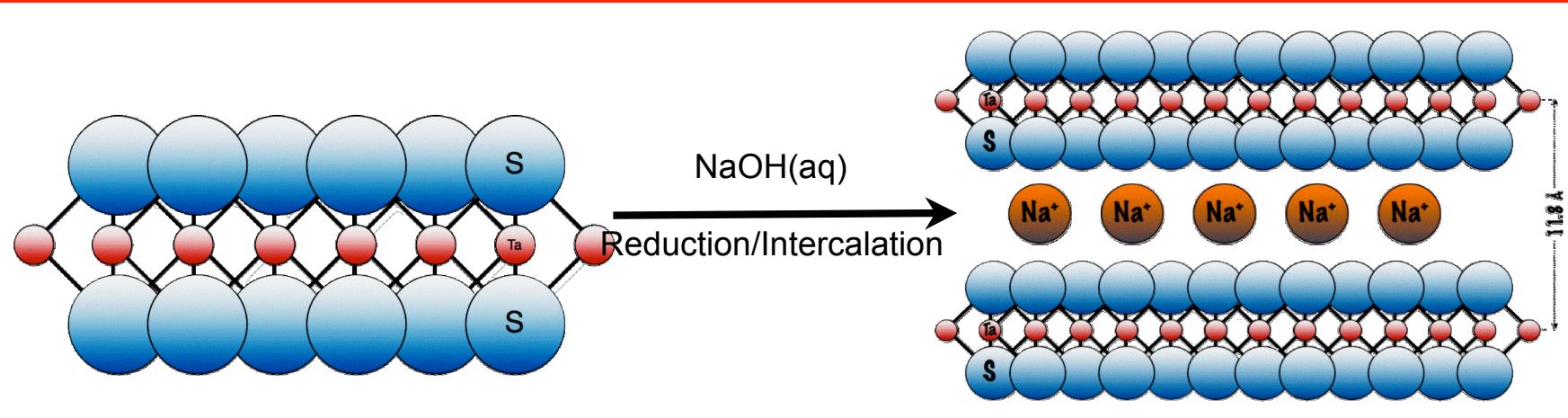
Magnetic Characterization

CATIONIC Mn_4 MOLECULES



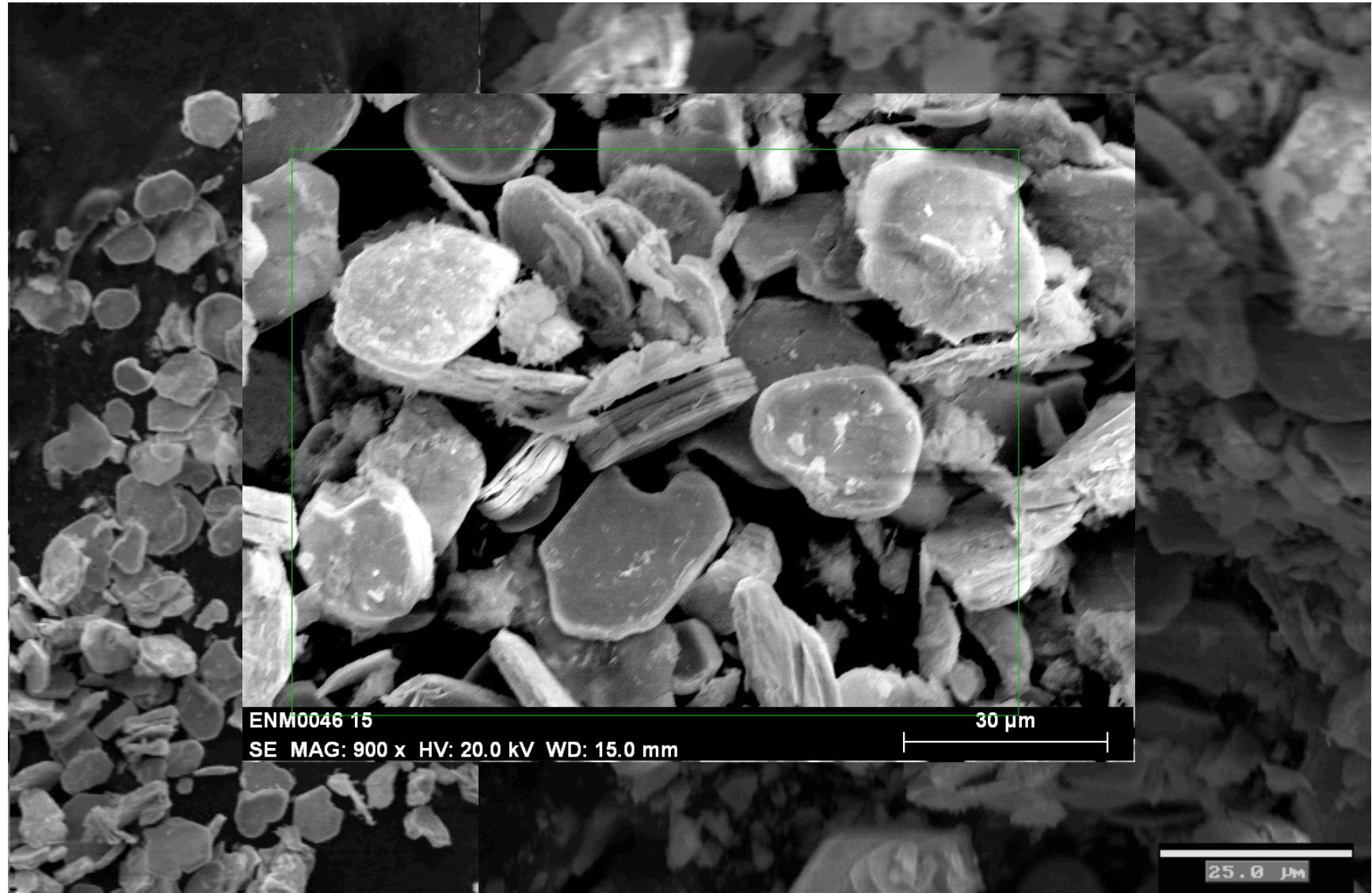
AC magnetic susceptibility at 1, 10, 332 and 1000 Hz.

A layered superconductor



Mn₄ intercalated in a layered superconductor

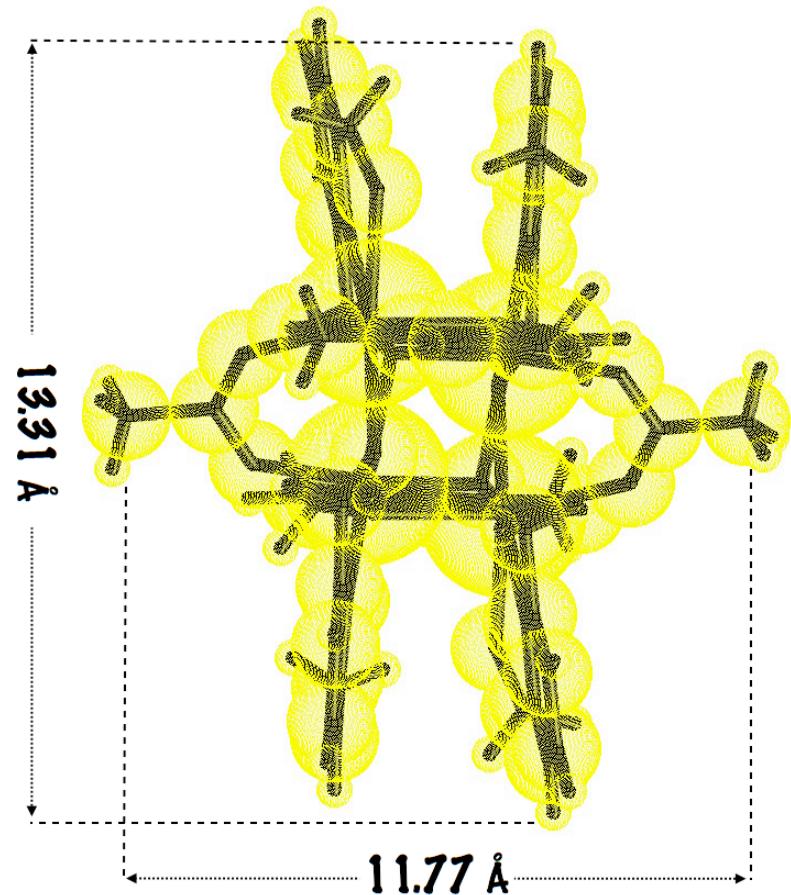
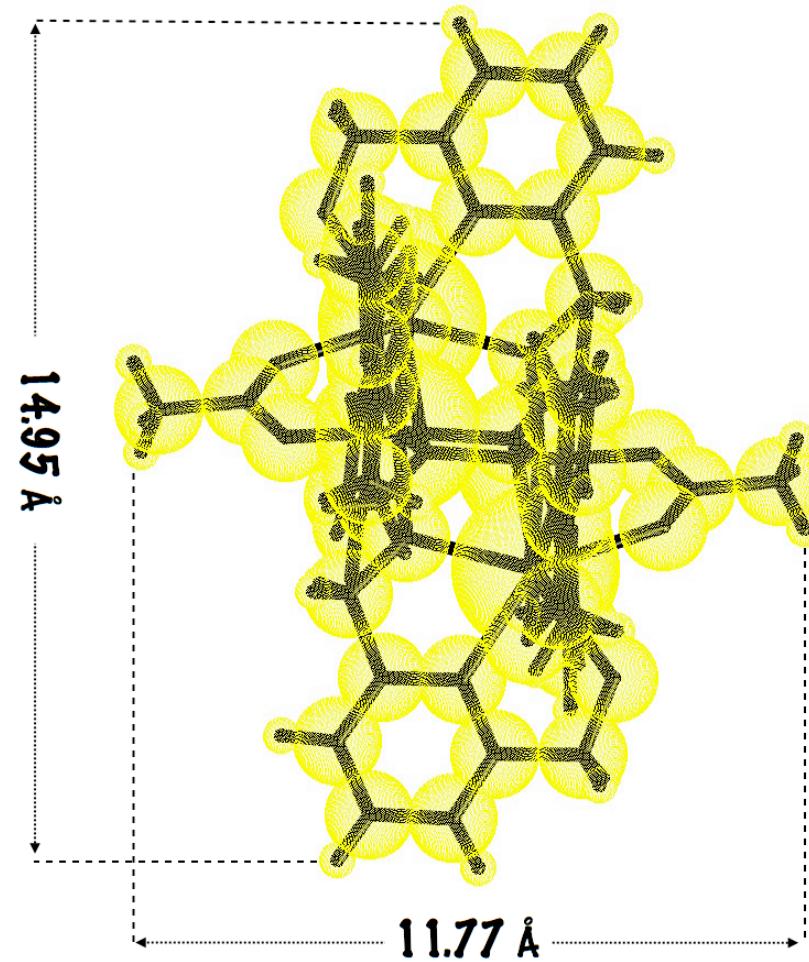
Morphology



Pictures aquired in a Hitachi S-4100 Scanning Electron Microscope

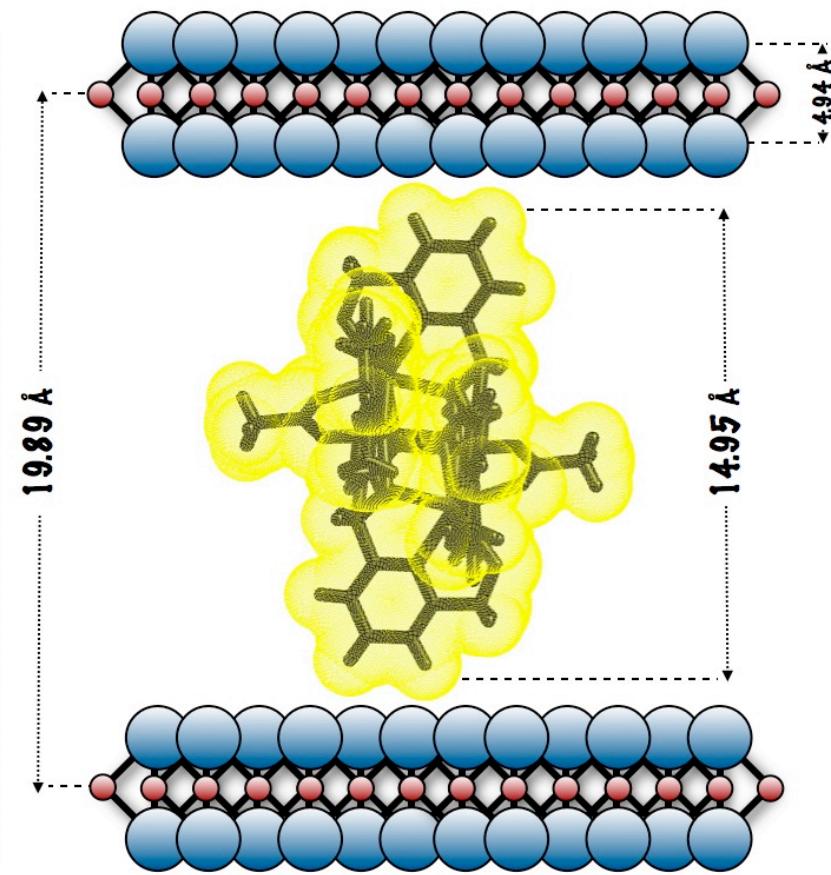
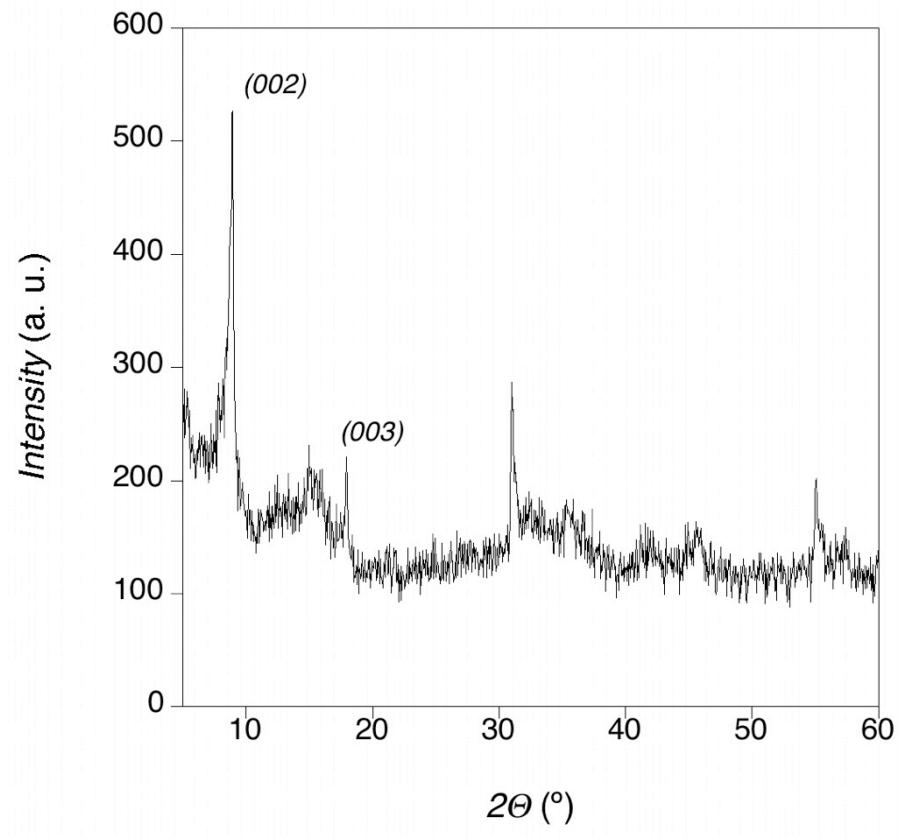
Mn₄ intercalated in a layered superconductor

Dimensions



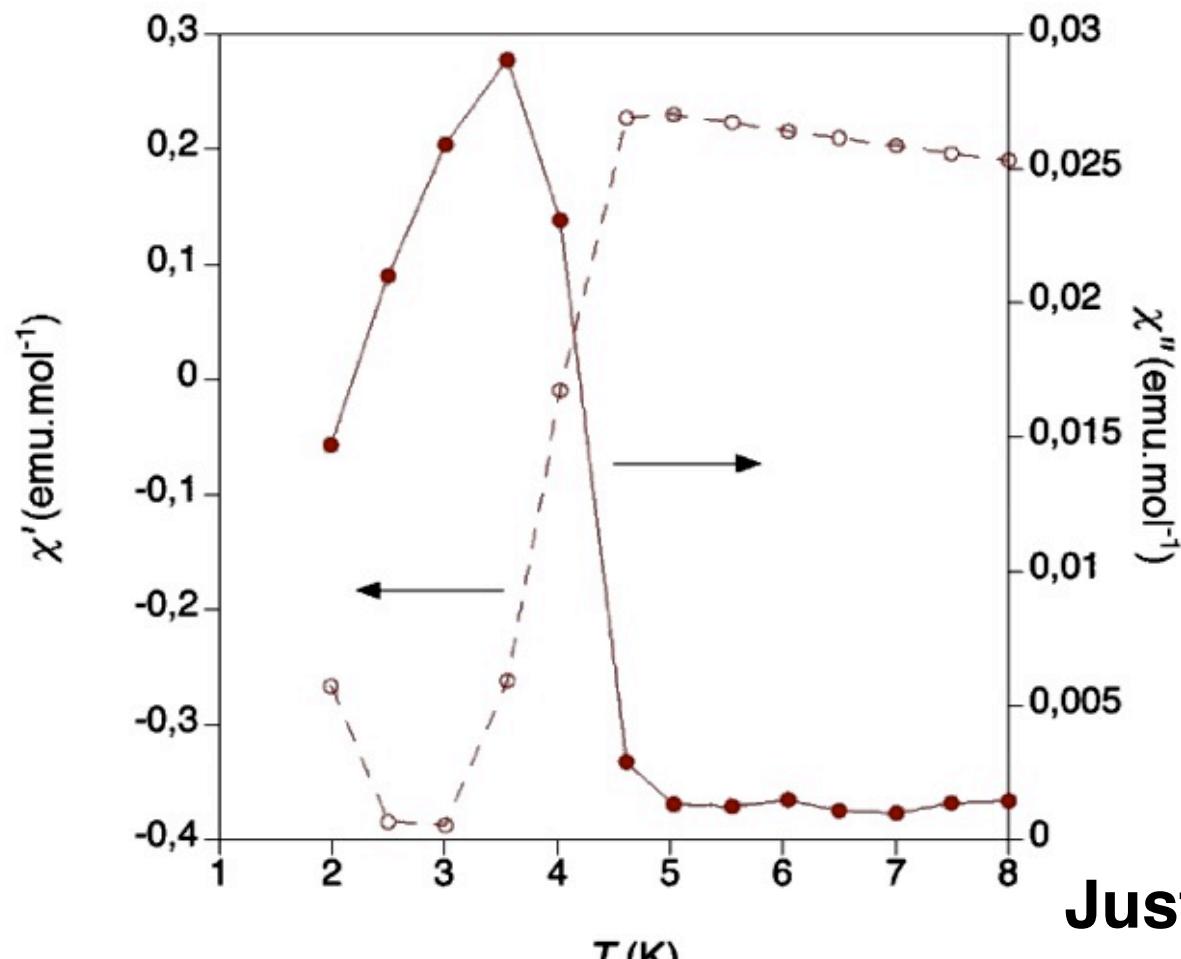
Structure

Mn₄ intercalated in a layered superconductor



Physical Characterization

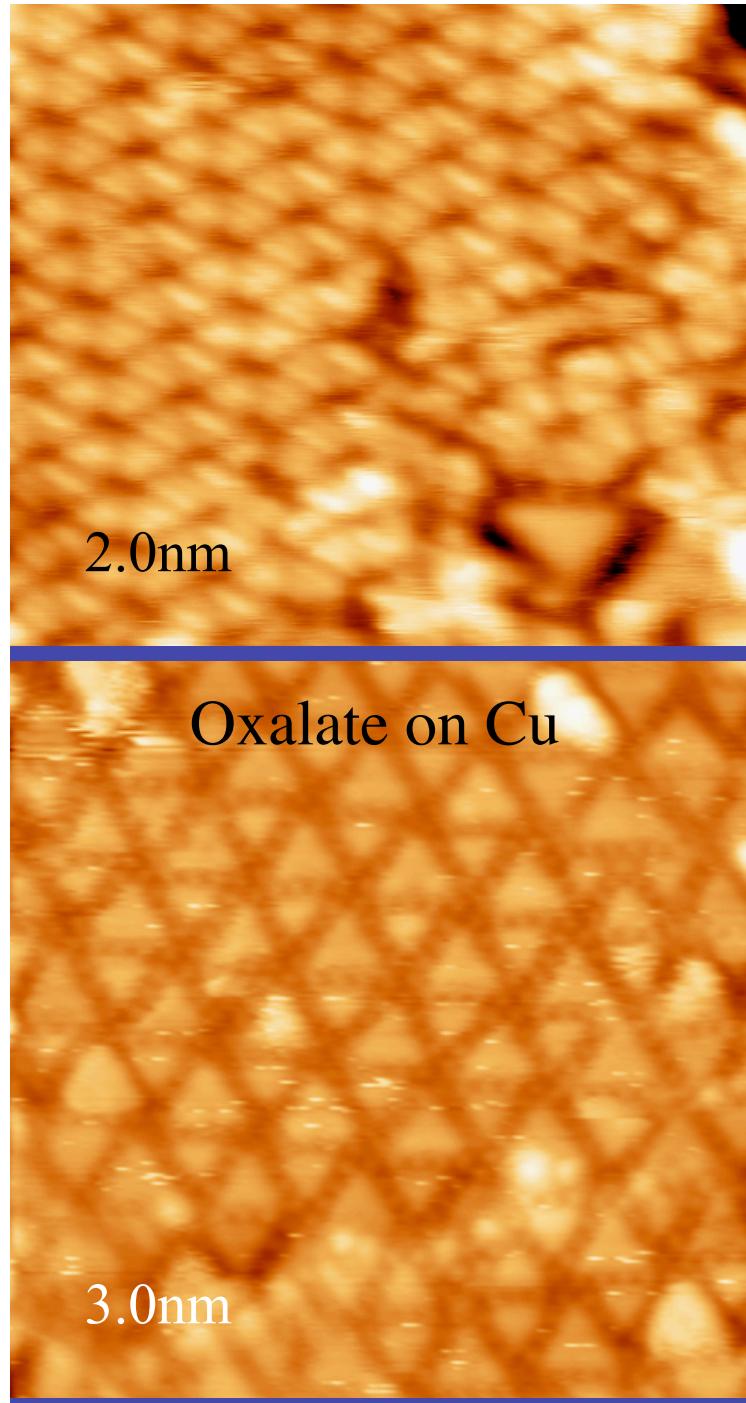
Mn₄ intercalated in a layered superconductor



Just coexistence or... interaction

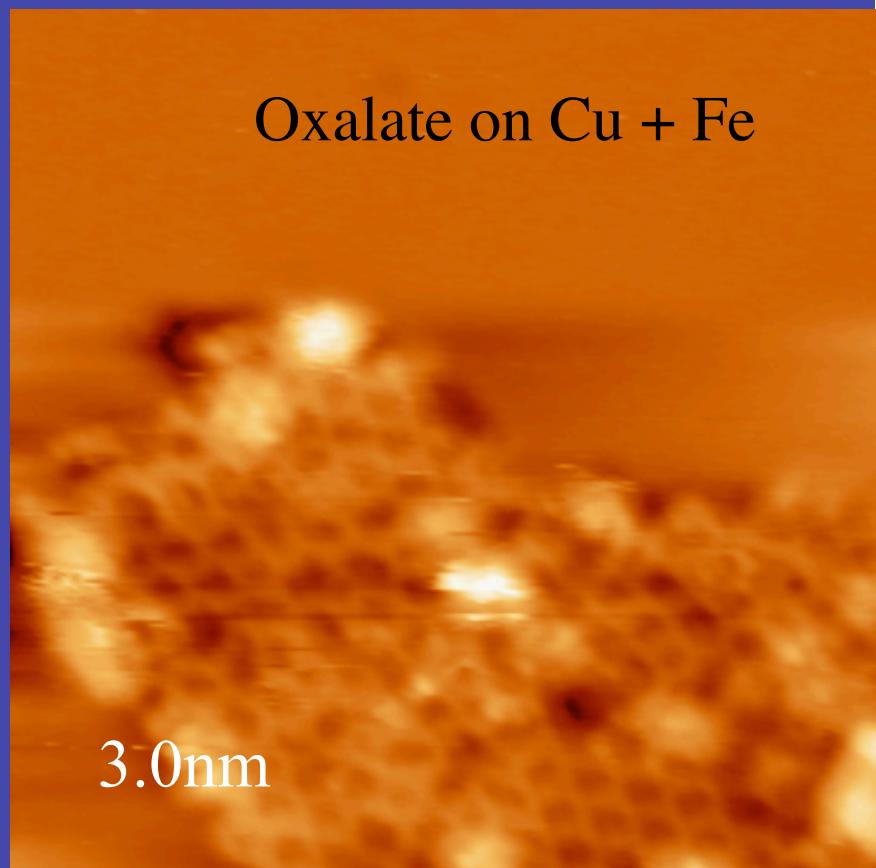
AC magnetic susceptibility of Mn₄-TaS₂ at 110 Hz.

The energy barrier is reduced by 1/2 (from 24 to 12 K): Screening of dipolar interactions by the superconductor ?????



MAGNETIC MONOLAYERS

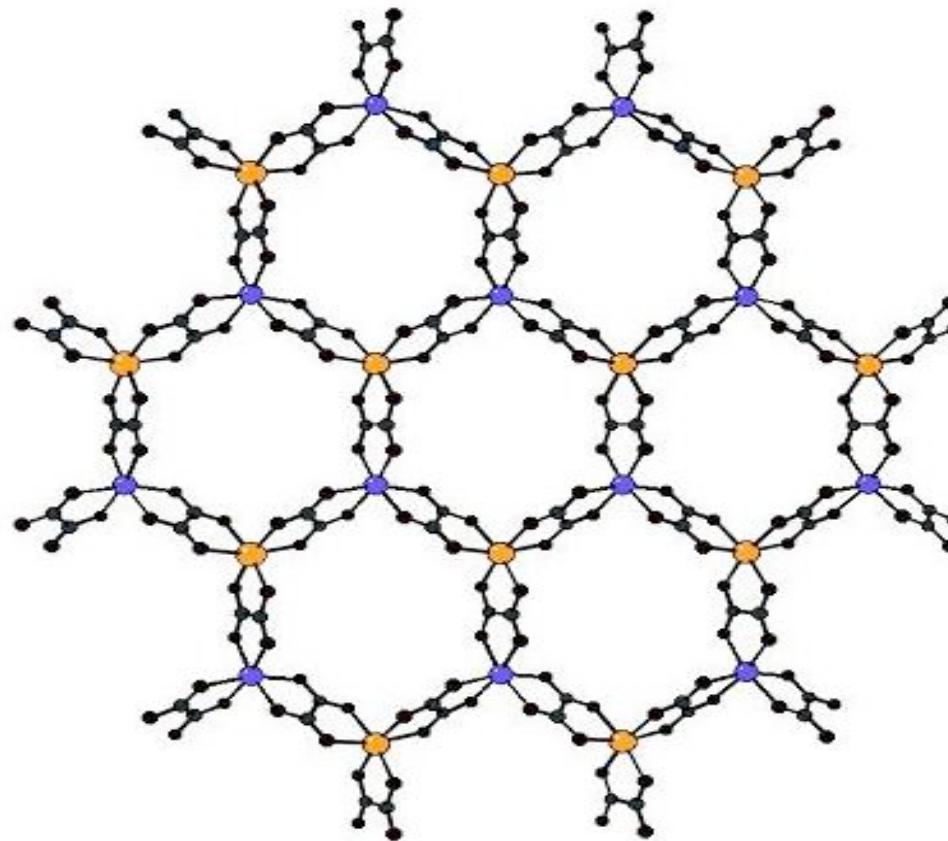
Oxalates on surfaces



Collaboration with R. Miranda

BIMETALLIC OXALATE LAYERS

$A[M^{II}M^{III}(ox)_3]$



2D